

Mechanical pretreatment of waste paper for biogas production

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Abstract

In the anaerobic digestion of lignocellulosic materials such as waste paper, the accessibility of microorganisms to the fermentable sugars is restricted by their complex structure. A mechanical pretreatment with a Hollander beater was assessed in order to reduce the biomass particle size and to increase the feedstock' specific surface area available to the microorganisms, and therefore improve the biogas yield. Pretreatment of paper waste for 60 min improves the methane yield by 21%, from a value of 210 ml/gVS correspondent to untreated paper waste to 254 ml/gVS. 30 min pretreatment have no significant effect on the methane yield. A response surface methodology was used in order to evaluate the effect of the beating time and feedstock/inoculum ratio on the methane yield. An optimum methane yield of 253 ml/gVS resulted at 55 min beating pretreatment and a F/I ratio of 0.3.

Keywords: renewable energy, biogas, biomass, waste paper, mechanical pretreatment, anaerobic digestion

Abbreviations: AD, anaerobic digestion; ANOVA, analysis of variance; BT, beating time; CCD, central composite design; CHPP, combustion and heat power plant; F/I, feedstock/inoculum; KDP, potassium dihydrogen phosphate; MC, moisture content; MSW, municipal solid waste; RSM, response surface methodology; TS, total solids; VFA, volatile fatty acids; VS, volatile solids.

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29 1 INTRODUCTION

30 Paper and cardboard are a heterogeneous mixture of plant material such as cellulose, hemi-cellulose,
31 lignin and filling material such as clay and calcium carbonate. Chemical additives (i.e. rosin, alum, starch)
32 are added to modify quality of the material and its properties such as brightness, opacity, or glossiness.
33 Cellulose is the major biodegradable fraction of waste paper but lignin is a recalcitrant compound for
34 anaerobic digestion and reduces the bioavailability of the cellulose (Zheng et al., 2014). Residual
35 contents of chemicals used during processing, such as talc or sodium silicate may still be found in the
36 paper product and consequently also in waste paper (European IPPC Bureau, 2013; Gran, 2001;
37 Villanueva and Eder, 2011). In Europe the per capita consumption of paper and board was 137 kg in
38 2012, in United Kingdom the total consumption was 1,0095,000 tonnes (Magnaghi, 2014). The biggest
39 source of recovered paper is industry and businesses with the 52% of the total, this covers the converting
40 losses (cuttings and shavings) and returns of unsold newspapers and magazines. Around 10% comes
41 from offices, and the remaining 38% from households (The Bureau of International Recycling, n.d.).

42 In United Kingdom, waste paper is mainly disposal to landfill, becoming the major contributor to municipal
43 solid waste by both volume (reaching the 50%) and weight. The space for approved and licensed landfills
44 will run out by 2020 (Infrastructure and Projects Authority, 2016). This fact alongside with leaching and
45 greenhouse gases emissions from the landfills requires other ways of waste paper treatment. A major
46 way of paper waste recycling is in paper mills, but some other uses are being investigated such as
47 construction materials (Folorunso and Anyata, 2007; Sutcu et al., 2014), animal bedding (Ward et al.,
48 2000), composting (Alvarez et al., 2009) or as a fuel (Brummer et al., 2014; Li and Liu, 2000). Many
49 studies have been carried out about the anaerobic digestion of pulp and paper sludge (Lin et al., 2011;
50 Meyer and Edwards, 2014; Priadi et al., 2014; Szeinbaum, 2009) and municipal solid waste (MSW)
51 (partially composed by paper and cardboard) (Kayhanian and Rich, 1995; Lo et al., 2012). In anaerobic
52 digestion, hydrolysis appears to be the rate-limiting step highly particulate waste, like paper waste
53 (Palenzuela Rollón, 1999). During this stage the degradation of cellulose and recalcitrant compounds like
54 lignin occurs. Hydrolysis depends on multiple factors such as the particle sizes of the substrate, pH and
55 enzymatic permeability of the substrate's membranes (Montingelli et al., 2015; Silvia Tedesco et al.,
56 2014). The availability of the substrates for the enzymatic attack will be achieved through the increment of
57 the specific surface area and breakdown the crystalline structure. In recent years different technologies
58 for biomass pretreatment have been developed in order to increase the availability of substrate for
59 anaerobic digestion (Kumar et al., 2009; Menind and Normak, 2008). Breaking down lignin, disrupting the
60 crystalline structure of cellulose and increasing its surface can be attained by pre-treatment methods, so
61 that micro-organisms can more easily access the cellulose (Behera et al., 2014). Although performing
62 pre-treatment makes the process more complicated and expensive, it can improve the process efficiency

63 and reduce the whole cost so that a positive energy balance can be obtained compared with non-pre-
64 treated biomass (Hendriks and Zeeman, 2009; Rodriguez et al., 2015). Mechanical, ultrasounds,
65 microwave, thermal, chemical and biological are the main pretreatment methods applied (C. Rodriguez et
66 al., 2016; Cristina Rodriguez et al., 2016). Mechanical techniques are the most efficient pretreatment for
67 biomass with complex structures, milling sisal fibres up to 2mm of particle size improved the methane
68 yield by 23% (Mshandete et al., 2006), the use of two commercially available heavy plates, resulted in
69 25% increase in the methane yield of ensiled meadow grass compared to the untreated feedstock
70 (Tsapekos et al., 2015). Mechanically milled rice straw achieved a 85% extra methane than untreated
71 material (Sasaki et al., 2016). Beating pretreatment with a Hollander beater for 15 min improved the
72 biogas yield of macroalgae *Laminaria* sp. by 36% and *Ascophyllum nodosum* by 26% (M.E. Montingelli
73 et al., 2016; Montingelli et al., 2017).

74 Only two pretreatment techniques have been reported in the literature to improve the biodegradability of
75 paper and cardboard: mechanical and biological. The mechanical pretreatment consisted in shred the
76 paper and cardboard fraction of municipal solid waste before anaerobic digestion but it has no significant
77 effect on biogas yields and on kinetics (Pommier et al., 2010). Better results were obtained when filter
78 paper, waste paper, newspaper and cardboard were pretreated with a thermophilic cellulose-degrading
79 consortium (MC1). After 55 days of anaerobic digestion, the methane yield of pretreated filter paper,
80 waste paper, newspaper and cardboard were 277, 287, 192, and 231 ml CH₄/gVS respectively, with
81 corresponding increases of 33%, 34%, 156%, and 141% with respect to the untreated materials (Yuan et
82 al., 2012). However biological pretreatments are slow processes, usually with residence times of 10–14
83 days, they require large amount of space and each feedstock requires a specific enzyme, forcing to study
84 an enzyme-substrate specificity (Rodriguez et al., 2015).

85 This paper investigates the improvements provided by a Hollander beater pretreatment. This technique is
86 based on the same 'comminution' concept proposed by all other mechanical treatments. The Hollander
87 beater has never been used as mechanical pretreatment machine on paper wastes. Seeing that this
88 proposed pretreatment has already proved its effectiveness when applied to seaweed biomass with an
89 improvement in biogas yield up to 20% (S. Tedesco et al., 2014; Tedesco et al., 2013), in this study it
90 has been applied to paper wastes in batch mode.

91 **2 MATERIALS AND METHODS**

92 **2.1 Feedstock and inoculum**

93 Waste paper was collected from recycle bins at the School of Computing and Engineering at the
94 University of West of Scotland (UWS) in Paisley, Scotland (Figure 1). This paper was mostly one side
95 printed and was cut by a shredder Fellowes Powershred C-320 in 0.6 x 29.7 cm pieces. The sludge used
96 as inoculum was provided by the Energen Biogas Plant (Cumbernauld, Scotland), and stored in a fridge
97 at 4°C. The plant uses food and food processing residues as a feedstock, the process is carried out under
98 thermophilic conditions.



99

100 **Figure 1.** Shredded paper (before pretreatment) and paper pulp (after pretreatment).

101 The total solids (TS) and volatile solids (VS) of the waste paper were calculated by duplicate and were
102 obtained by submitting random samples of pretreated waste paper at 105°C (for TS) and 550°C (for VS)
103 until constant weight. The sludge's characterization is provided by the supplier. The methane production
104 is provided in terms of volume per gram of VS (ml/gVS). The characterization of the paper and the sludge
105 is detailed in Table 1.

106 **Table 1.** Waste paper and sludge characterization.

Parameters	Sludge	Untreated paper	30 min pret. paper	60 min pret. paper
Total Solids (%)	5	95	3	3
Volatile Solids (% of TS)	63	99	97	97
Ash content (% of TS)	37	1	3	3

107

108 **2.2 Hollander beater pretreatment**

109 The machine consists of a modified Hollander beater (Figure 2). This beater is normally used in the paper
110 industry (Lumiainen, 2000). Most of the mechanical pretreatments can be done in existing facilities
111 previously used for other purposes and other materials. This is a great advantage as these facilities only
112 need with minor changes or adjustments in order to use them in the biomass pretreatment process.
113 The feedstock is exposed to the shear action in the beater, blades and grooves exercise a cutting action
114 while the high pressure and speed reached under the drum beats the mixture. The biomass should be
115 soaked prior its treatment in the beater, in the case of paper as it is a thin and absorbent material, it can
116 be soaked for one hour (Cerde, 2008; Osorio, 2010). The capacity of the beater is about 1 kg of dry
117 biomass, but this can vary depending upon the type of feedstock.



118

119 **Figure 2.** Hollander beater in operation with waste paper.

120 Samples were taken at 30 and 60 min of beating pretreatment. The samples were taken from the bend
121 before the bladed drum in the middle of both the width and height of the channel to take the most
122 representative sample.

123 **2.3 Experimental set-up**

124 The bioreactors consisted of 500 ml Erlenmeyer flasks with working volume of 400 ml connected through
125 a system of valves and plastic pipes to airtight Linde PLASTIGAS bags for biogas collection (Figure 3).
126 To clear up any trace of oxygen from the system and preserve the anaerobic conditions, nitrogen was
127 flushed into the reactors headspace during 5 min and then removed. This operation was done three
128 times. The reactors were placed in a water-bath to keep a mesophilic temperature of 37°C.



129

130 **Figure 3.** Anaerobic reactors with biogas collection systems.

131 Reactors were fed with a fixed amount of 200ml of sludge (inoculum), while different quantities of pulp
132 (beated paper) were required to have different F/I ratios as (0.3, 0.5 and 0.7). The pH was adjusted to
133 7.00 ± 0.15 with potassium dihydrogen phosphate (KDP) as a buffer solution. The reactors corresponding
134 to the untreated samples were fed with shredded paper. In order to assess the inoculum contribution to
135 the methane production, control batches were prepared in the same way except for the paper addition.
136 Flasks were daily shaken during the process in order to favour the degasification of the substrate and the
137 contact between the biomass and the inoculum. Each test was conducted by duplicated, and the average
138 results were reported in this paper.

139 For gas volume measurement was used a graduated upside-down cylinder connected to a bubbling flask
140 in order to maintain the necessary oxygen-free conditions and avoid air infiltrations. A gas analyser
141 (Dräger X-Am 7000.) was used to determine the biochemical composition of the obtained biogas The
142 digestion was stopped according to (VDI-Gesellschaft Energietechnik, 2006) when the daily biogas

143 production rate was found to be less than 1% of the overall volume produced. The biogas volumes are
144 given for a dry gas in standard conditions of temperature (0°C) and pressure (1 atm). As the biogas
145 produced is saturated with water vapour, the water content was removed from the results as well. The
146 inoculum contribution to biogas production was never higher than 10%

147 2.4 Design of experiments

148 The experiment was planned according to a response surface methodology (RSM) for two factors,
149 beating time and F/I ratio with three levels; the response was the biogas production per g of volatile solids
150 (VS). RSM is characterized by high adherence to the experimental data describing the reality being
151 studied, the method captures accurate efficient approximations for accurate data from numerical or
152 practical experiments at discrete data points in the design space (Benyounis and Olabi, 2008). Moreover,
153 RSM methods are able to exhibit the factor contributions from the coefficients in the regression model and
154 identify the insignificant factors and thereby can reduce the complexity of the problem (Montingelli et al.,
155 2017). Response surface methodology consists of a group of mathematical and statistical techniques
156 used in the development of an adequate functional relationship between a response of interest, y , and a
157 number of associated control (or input) variables denoted by x_1, x_2, \dots, x_k . Usually, a second order
158 polynomial as shown in Equation 1 is used in RSM to describe the true functional relationship between
159 the independent variables and the response surface:

$$160 \quad Y = b_0 + \sum b_i x_i + \sum b_{ii} x_{ii}^2 + \sum b_{ij} x_i x_j \quad (1)$$

162 where the values of the model coefficients b_0 , b_i , b_{ii} and b_{ij} are estimated using regression analysis (Maria
163 E. Montingelli et al., 2016). In this study, the RSM was applied through a central composite design (CCD)
164 to fit a model by least squares technique. CCD is a factorial or fractional factorial design with centre
165 points, augmented with a group of axial points (also called star points) that led to curvature estimation. It
166 can be used a central to efficiently estimate first- and second-order terms and model a response variable
167 with curvature by adding centre and axial points to a previously-done factorial design (Ahmadi et al.,
168 2005; Ryan, 2007; Vining and Kowalski, 2010).

169 The arrangement of CCD as shown in Table 2 was in such a way that allows the development of the
170 appropriate second order polynomial equation.

171

172

173

Table 2. Arrangement of the CCD for the two independent variables used in the present study

Experiment n°	Variable levels/coded values	
	Beating time (x_1)	Feedstock/Inoculum ratio (x_2)
1	-1	-1
2	0	-1
3	1	-1
4	-1	0
5	0	0
6	1	0
7	-1	1
8	0	1
9	1	1

174

175 Factor levels and independent input variables are respectively 0, 30 and 60 minutes for the beating time
176 (BT) and 0.3, 0.5 and 0.7 for feedstock/inoculum ratio (F/I). Level 0 of factor BT represents untreated paper
177 waste.

178 The adequacy of the models is tested through the analysis of variance (ANOVA). The statistical significance
179 of the models and of each term is examined using the sequential F-test and lack-of-fit test. If the Prob. > F
180 of the model and of each term in the model does not exceed the level of significance (in this case $\alpha = 0.05$)
181 then the model may be considered adequate within the confidence interval of $(1 - \alpha)$. An adequate model
182 means that the reduced model has successfully passed all the required statistical tests and can be used to
183 predict the responses or to optimize the process. The values of R^2 , adjusted- R^2 , predicted- R^2 , lack of fit
184 and adequate precision of models are obtained to check the quality of the suggested polynomial. The
185 statistical study was performed using the Design Expert software version 9.

186

187 **2.5 Methane production rate**

188 A first order model (Equation 2) was used to describe the progress of cumulative methane production
189 obtained from the batch experiments (Jokela et al., 2005; Lin et al., 2011).

190
$$B(t) = B_0(1 - e^{-kt}) \quad (2)$$

191 where B (t) is the cumulative methane production (ml/gVS), B₀ is the maximum methane production
 192 (ml/gVS), k is the methane production rate constant (d⁻¹), and t is the time (d). Biodegradability results
 193 were compared after a significance statistical analysis by using analysis of variance (ANOVA) for a single
 194 factor. Statistical significance was established at p < 0.05 level.

195 **3 RESULTS AND DISCUSSION**

196 **3.1 Methane production**

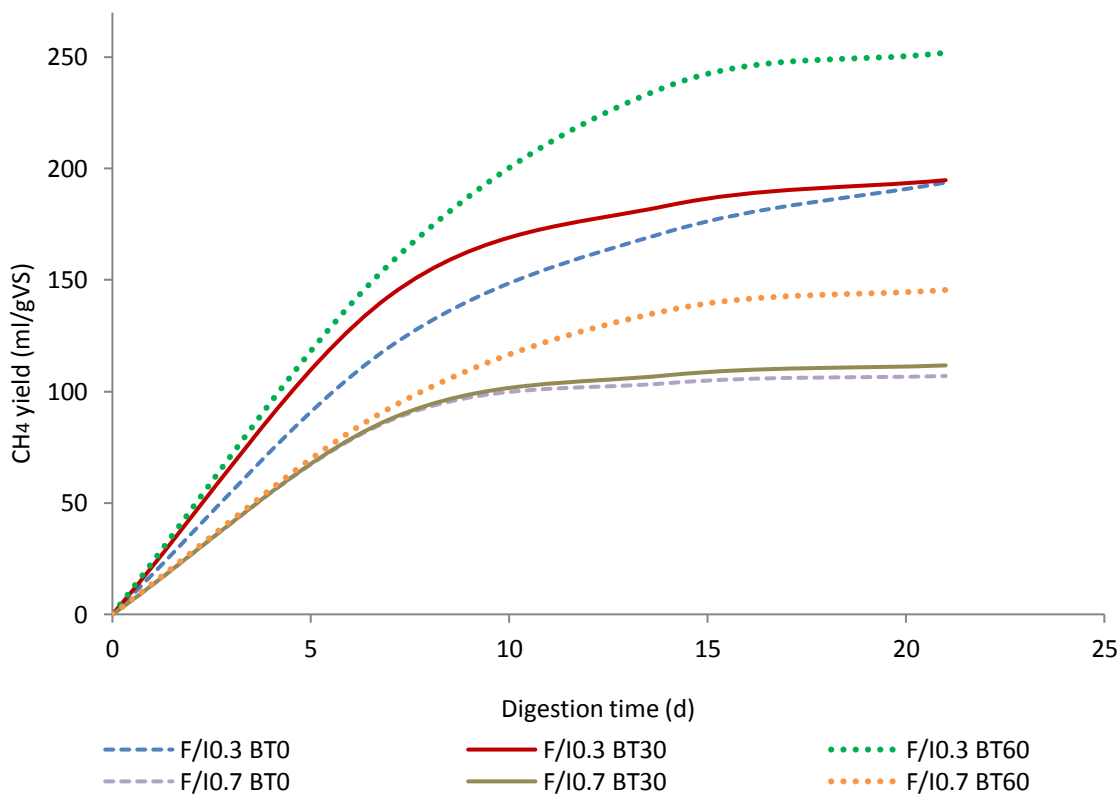
197 The present means and standard deviations of performed experiments are shown in Table 3.

198 **Table 3.** Experimental results obtained at the end of the biodegradability tests

Ratio F/l	Beating time (min)	Methane production (ml/gVS)	k (d ⁻¹)	pH
0.3	0	210±8	0.12±0.01	7.13±0.07
	30	199±7	0.18±0.01	6.65±0.14
	60	253±12	0.14±0.01	7.04±0.08
0.5	0	132±7	0.20±0.01	7.05±0.06
	30	120±9	0.24±0.01	6.70±0.20
	60	215±9	0.10±0.01	6.98±0.10
0.7	0	107±4	0.24±0.01	6.89±0.27
	30	112±12	0.21±0.01	6.98±0.06
	60	175±11	0.09±0.01	7.03±0.04

199
 200 The methane yield decreased with increased ratio F/l for all pretreatment times. For the untreated paper,
 201 the methane yield decreased by 37% from 210 ml/gVS correspondent to ratio 0.3 to 132 ml/gVS for ratio
 202 0.5. For 60 min pretreated paper, the methane yield at ratio 0.7 was 175 ml/gVS, which was a 31% less
 203 than for a ratio of 0.3. F/l ratio affects the methane production rate, the consumption of VFAs and the
 204 methane yield. To achieve maximum methane yields and a stable process, the F/l ratio is a crucial
 205 parameter and should be lower than 1 in terms of VS. An optimum F/l ratio ensures the presence of the
 206 groups of microorganisms required for the complete anaerobic digestion (Ali Shah et al., 2014). Knowing

207 the optimum F/I ratio allows a better exploitation of the feedstock. Feeding the reactor with high quantities
 208 of biomass that the inoculum is not able to process lead to a loss of feedstock, that is not digested. Methane
 209 yield for untreated macroalgal at F/I 0.7 was 49% lower than for F/I 0.3, this stands for half of the biomass
 210 not digested, when the biomass is beaten for 60min, the decreased in methane yield from 0.3 to 0.7 F/I is
 211 30%, this means, 30% of the digested biomass at low F/I ratio was not digested at F/I 0.7. Similar results
 212 were achieved on sunflower oil cake anaerobic degradation with the methane yield decreasing considerably
 213 from 227 to 107 ml/gVS when the F/I increased from 0.33 to 2, showing a marked influence of this parameter
 214 on methane yield (Raposo et al., 2008). On municipal solid waste degradation, the optimum F/I ratio were
 215 the lowest value tested (Boulanger et al., 2012), a maximized biogas production from cattle manure was
 216 obtained at a minimum F/I tested (Johari and Widiassa, 2012). However, in other cases the F/I ratio had
 217 minor effect in the methane yield (Eskicioglu and Ghorbani, 2011; González-Fernández and García-Encina,
 218 2009). The influence of the F/I ratio on the methane yield depends also in the F/I ratio range tested; near
 219 the optimum F/I ratio the influence will be less noticeable.



220

221

Figure 4. Methane production for low (0.3) and high (0.7) F/I

222 At the early stages of the degradation (day 7) for a F/I ratio of 0.3, the methane yield from 30 min beated
223 samples is 13% higher than for untreated material (Figure 4). 60 min beated paper produced 43% more
224 methane than untreated biomass on the same day. These improvements continued in day 14 of
225 digestion, when 30 min pretreatment improved the methane yield by 8% and 60 min pretreatment by
226 26%. The methane yields improvements on day 14 are roughly the half of improvements in day 7, and at
227 the end of the digestion only 60 min pretreatment achieved a positive effect on the methane yield. Higher
228 methane production rate constants were achieved for 30 min beating pretreatment at F/I ratios of 0.3 and
229 0.5, however the final methane production is lower than for 60 min pretreatment. This trend can be
230 explained due to that the first step of lignocellulosic materials degradation is hydrolysis of the cellulose. It
231 takes place at the surface of the cellulose fibers; therefore, more beated samples achieved more specific
232 surface area accelerating the hydrolysis. The low first order constants and high final methane productions
233 achieved for 60min beated samples shows that contrary to expected, the hydrolysis of cellulose is maybe
234 not the limiting step of the waste paper degradation process. agreed well with Keymer et al. (Keymer et
235 al., 2013), who noticed that the high pressure thermal hydrolysis pretreatment had no effect on the
236 methane production rate but significantly improved the final methane yield of *Scenedesmus* microalgae;
237 similar results were obtained with olive mill solid waste, where co-digestion with *D. salina* improved the
238 total methane production but had negative effect on the initial degradation rate (Fernández-Rodríguez et
239 al., 2014).

240 At the end of the degradation, the methane yield for a ratio F/I of 0.3 decreased by 5% when the paper
241 waste was beated for 30 min, such percentage is not statistically significant when compared with the
242 batch duplicates, so it can be concluded that 30 min pretreatment at 0.3 F/I ratio have no effect on the
243 methane yield . When the pretreatment time was increased to 60 min, the methane yield increased by
244 21% from 210 ml/gVS correspondent to the untreated paper to 253 ml/gVS. The present result from non
245 beated paper is consistent with the data from Eleazer et al (Eleazer et al., 1997), where waste paper yield
246 220 mlCH₄/gVS. A short beating time (30min) increases the methane production rate however; the final
247 methane yield is much lower compared to 60min beating pretreatment. The pretreatment seems start to
248 be effective after 60 min being that methane production for 60 min treatment is higher than for both
249 untreated and 30 min treated paper.

250

251 **3.2 Process modelling**

252 The experimental factors, F/I and BT were checked in three levels. Beating time varies between 0 and 60
 253 minutes and ratio feedstock/inoculum varies between 0.3 and 0.7. The response was the methane
 254 production given in ml per g of volatile solids (ml/gVS). Parameters and results are presented in Table 4.

255 **Table 4.** Experimental factors and response in arrangement for the CCD used in the present study

Experiment n°	Experimental factors		Response
	Beating time (min)	Ratio F/I	Methane yield (ml/gVS)
1	0	0.3	210
2	0	0.5	132
3	0	0.7	107
4	30	0.3	199
5	30	0.5	120
6	30	0.5	120
7	30	0.7	112
8	60	0.3	253
9	60	0.5	215
10	60	0.7	175

256 For the optimization through the RSM of the methane yield, the model F-value of 36.43 implies the model
 257 is significant. The model terms of $R^2 = 0.9785$, adjusted- $R^2 = 0.9517$, predicted- $R^2 = 0.8127$, all these values
 258 are very close to 1 and so indicate the adopted model is adequate. The final mathematical model associated
 259 to the response in terms of actual factors in Equation 3 and the ANOVA test is shown in Table 5.

260
$$\begin{aligned} \text{Methane yield} = & 401.86 - 2.12BT - 812.85F / I \\ & + 1.02BT * F / I + 0.04BT^2 + 559.70(F/I)^2 \end{aligned} \quad (3)$$

261

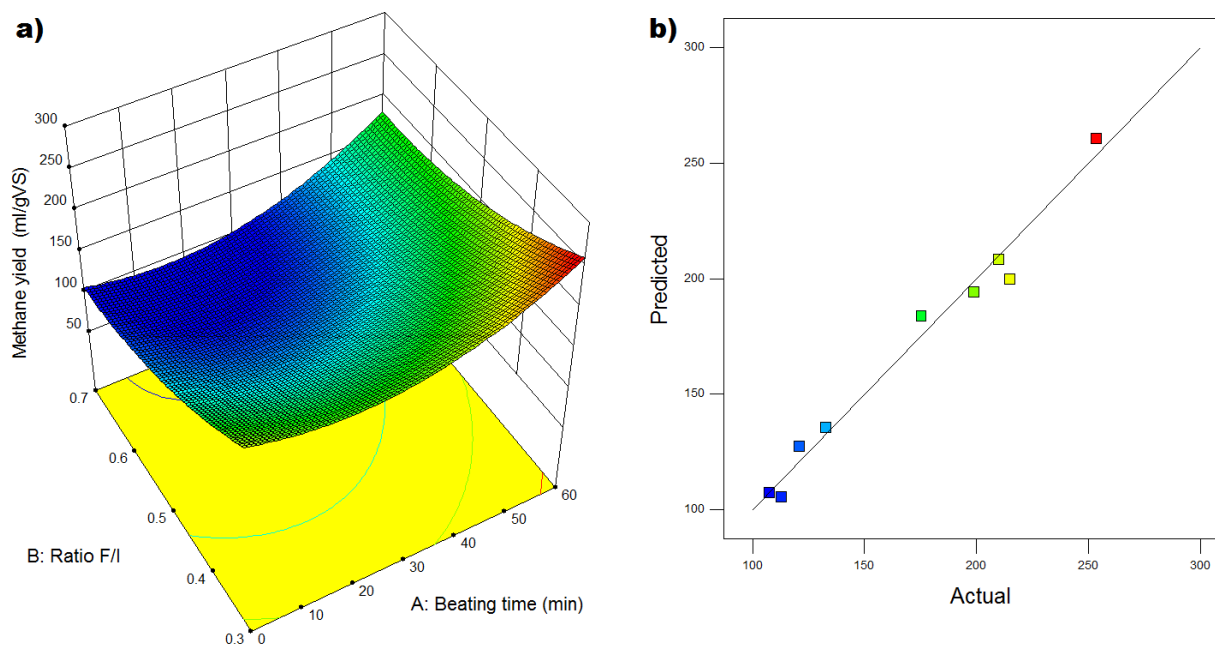
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Table 5. ANOVA test from response surface design for methane yield.

Source	Sum of Squares	Mean Square	F Value	p-value	Prob > F
Model	24086.82	4817.36	36.43	0.0020	
A-Beating time	6248.03	6248.03	47.25	0.0023	
B-F/I ratio	11869.52	11869.52	89.77	0.0007	
AB	151.39	151.39	1.14	0.3449	
A ²	3785.40	3785.40	28.63	0.0059	
B ²	1169.52	1169.52	8.85	0.0410	
Residual	528.90	132.22			
Cor Total	24615.72				

265 The response surface obtained from the model illustrated in Figure 5a shows that higher methane yields
 266 are obtained at high beating times and low F/I ratios. The predicted vs. actuals plot (Figure 5b) shows that
 267 these values were distribute near to a straight line and a satisfactory correlation between them is observed.
 268 This demonstrates that the model can be effectively applied for mechanical pretreatment with a Hollander
 269 beater for paper waste.



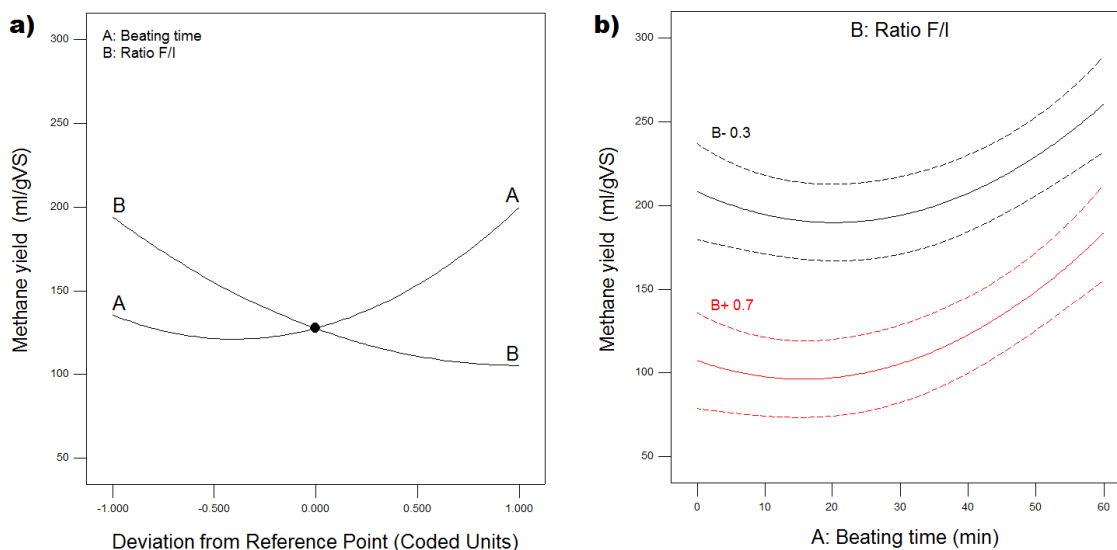
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271 **Figure 5.** Response surface plot in 3D for methane yield (a) and scatter diagram for methane yield (b).

272

273 The perturbation plot in Figure 6a shows how the methane yield is affected by the input variables beating
 274 time and F/I ratio, both variables have an exponential effect on the methane production. Increasing B (F/I
 275 ratio) the methane yield will decrease exponentially. The effect of the beating time is the opposite, methane
 276 yield increases exponentially with the pretreatment time. The effect of pretreatment has a similar behaviour
 277 at low and high F/I ratios (Figure 6b). For a F/I ratio of 0.7, the methane yield achieved a minimum around
 278 27 min of pretreatment, for ratio F/I of 0.3 the minimum is achieved at around 23 min.

279



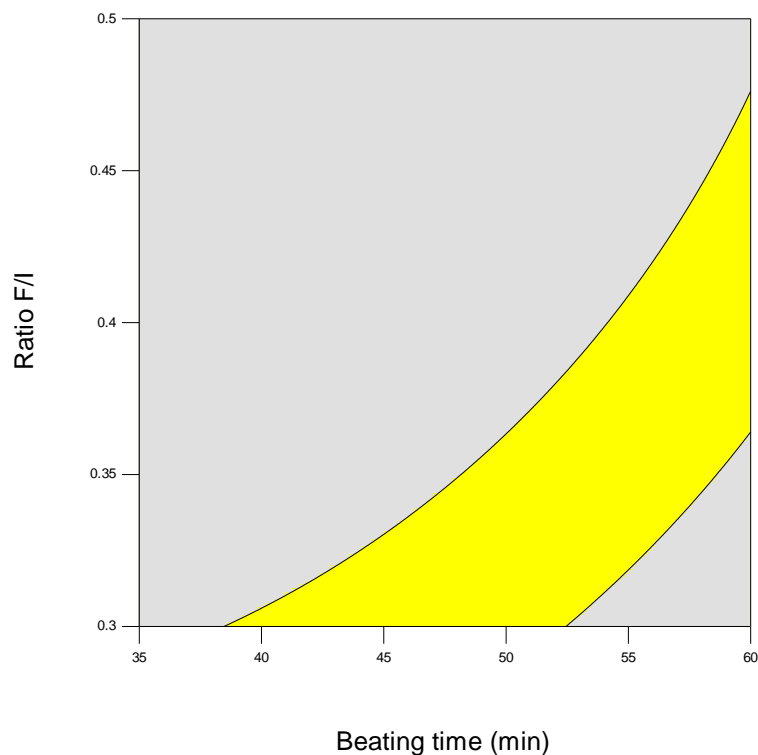
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281 **Figure 6.** Perturbation plot for methane yield (a) and interaction plot for methane yield (b).

282 3.3 Methane yield optimization

283 Based on the response surface model showed in Equation 3, which describes the effects of process
 284 parameters on the methane production, an optimization study was conducted using Design-expertV9
 285 software. The optimization criteria combine the productivity with the cost of the process, the methane yield
 286 was maximized with level 5 and beating time was minimized with level 1 while F/I ratio was permitted to
 287 vary in the same range as in Table 4.

288 The optimal methane yield of 245 ml/gVS from the numerical optimisation was found at BT= 55 min and F/I
 289 ratio= 0.3, allowing 17% extra methane when compared to the maximum methane production for untreated
 290 paper. The graphical optimization allows a selection of the optimum process parameters by means of visual
 291 inspection. The yellow areas on the overlay plot (Figure 7) that represent the values that meet the proposed
 292 criteria is delimited by the curves corresponding to the optimization criteria set by the authors.



293

294

Figure 7. Graphical optimization for maximizing methane yield while minimizing beating time.

295 Three confirmation experiments (including the optimal point) were carried out using new test conditions to

296 verify the adequacy of the models. The experimental conditions, the actual and predicted values and the

297 percentages of error are summarized in Table 7. Considering that anaerobic digestion is a biological

298 process highly influenced by the inoculum, the percentages of error are all within acceptable tolerances.

299

Table 6. Validation experiments

Experiment	Beating time (min)	Ratio F/I	Methane yield (ml/gVS)	
1	15	0.6	Actual	115
			Predicted	104
			Error (%)	9.33
2	45	0.4	Actual	179
			Predicted	190
			Error (%)	-6.41
3	55	0.3	Actual	245
			Predicted	260
			Error (%)	-60.4

300 4 CONCLUSIONS

301 The experimental work shows the methane yields obtained from the digestion of waste paper inoculated
302 with sludge from a biogas production plant. Pretreated waste paper with a Hollander beater for 60 min
303 improved the methane yield by 21%. 30 min pretreatment have no significant effect on the methane yield
304 even if the methane production rates increased. The highest methane yields were achieved at F/I ratio
305 0.3 for all pretreatment times.. An optimization study was performed to reduce the operating costs and
306 time associated to the pretreatment and maximizes the productivity. The aim is maximizing the methane
307 production while minimizing the pretreatment time. An optimized methane yield of 245 ml/gVS was
308 achieved for 55 min of beating pretreatment and a F/I ratio of 0.3 allowing 17% more methane than non
309 beaten waste paper.

310 The above findings summarize that mechanical pretreatment of waste paper in a Hollander beater led to
311 an increase in the final methane yields rather than the reaction kinetics. Further work will focus on
312 improving the anaerobic digestibility of mechanically pretreated waste paper through its codigestion with a
313 high nitrogen content feedstock as seaweed.

314

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