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4 **Stable Formamidinium Based Centimeter Long Two-Dimensional (2D) Lead Halide**
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6 **Perovskite Single Crystal for Long-Live Optoelectronic Applications.**
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4 **Abstract**
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7 Solution-processable 2D metal-halide perovskites are highly promising for cost-effective
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9 optoelectronic applications due to their intrinsic multi-quantum well structure. However, the lack
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11 of stability is still a major obstacle use of this class of materials in practical devices. Here, we
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13 demonstrate the stable optoelectronic properties using formamidinium (FA) based centimeter-long
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15 2D perovskite (BA)₂FAPb₂I₇ high-quality single-crystal controlled by the thickness of two
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17 perovskite layers. The large area single crystal exhibits good crystallinity, phase purity, and
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19 spectral uniformity. Moreover, the (BA)₂FAPb₂I₇ single crystal shows excellent stability at open
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21 atmospheric conditions when compared to methylammonium (MA) based (BA)₂MAPb₂I₇
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23 counterparts. The photodetectors fabricated using 2D perovskite single crystal on the rigid Si/SiO₂
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25 substrate reveal high photoresponsivity (R_{λ}), ~5 A/W, the fast response time, < 20 ms, specific
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27 detectivity (D^*), ~ 3.5×10^{11} Jones, and excellent durability under 488 nm laser illumination. The
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29 R_{λ} and D^* values were obtained from the (BA)₂FAPb₂I₇ single-crystal 25 times and three orders
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31 magnitudes, respectively higher than the (BA)₂MAPb₂I₇ single crystal. Additionally, the
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33 perovskite material on flexible polymer substrate reveals good photo-sensing properties in both
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35 bending and non-bending states.
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46 **Keywords:** 2D Perovskites, Ambient stability, Photoresponsivity, Specific detectivity, Flexible
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1. Introduction

ABX_3 (A = $CH_3NH_3^+$, B = Pb and X = Cl, Br and I) metal halide perovskites (MHPs) are currently being well explored due to outstanding solar cell performance with power conversion efficiency greater than 25% by solution process.^[1] High optical absorption coefficient^[2], low binding energy^[3], long carrier diffusion^[4], and tunable band gap^[5] make this class of materials promising in photodetector^[6], solar cell^[7], light-emitting diode (LED)^[8], and laser^[9] applications. However, chemical degradation due to poor ambient stability limits their use in practical applications. Ruddeldsen-Popper (RP) two-dimensional (2D) metal halide perovskites (2D-MHPs) represented by the generic formula $(A')_2A_{n-1}M_nX_{3n+1}$, where A' is a long-chain organic spacer, such as butyl ammonium (BA) cation ($C_4H_9NH_3^+$), 'A' is small organic methylammonium (MA) cation ($CH_3NH_3^+$), M is divalent metal ions (Pb^{2+}), X is halide ions (Cl^- , Br^- and I^-) and 'n' is integers (n=1,2,3,4...), drew much attention due to quantum confinement and better stability.^[10-13]

The 2D-MHP forms a self-assembled multi-quantum well (MQW) structure through the alternative stacking of metal-halide perovskite and long-chain organic spacers in the RP phase.^[10] The semiconducting perovskite slabs act as potential "wells," and the insulating cationic long-chain organic spacers act as potential "barrier" layers. The number of conductive slabs per unit cell determines quantum well thickness.^[10] Due to quantum confinement, 2D-MHPs exhibits stable exciton emission, high exciton binding energy, quantum yield, high carrier mobility, anisotropy, and tunable optoelectronic properties.^[10] Among existed 2D-MHPs, homologous $(BA)_2(MA)_n-1Pb_nX_{3n+1}$ with the range of 'n' values were widely explored in the form of large-area thin films and bulk single crystals for solar cells^[13,14], photodetector^[15], LEDs^[16,17], optical pumping^[18,19], and transistor^[20] applications.

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4 Although the 2D perovskites exhibit relatively better stability than their 3D counterparts,
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6 ^[10] atmospheric degradation persists as a severe problem in niching them into proper applications.
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8 Particularly, the perovskite architecture formed by the MA cation undergoes easy degradation in
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10 the ambient atmosphere due to structural instability caused by the low tolerance factor.^[21,22]
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12 Replacement of smaller ionic radius MA cation (~2.17 Å) with larger FA cation (~2.53 Å) was
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14 proved to potentially stabilize the crystal structure by sustaining the tolerance requirements which
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16 in turn enhanced the atmospheric and thermal degradation of the MHPs.^[22,23] However,
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18 applications of FA-based MHPs are mostly limited to 3D FAPbX₃ perovskites.^[23-25] The
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20 systematic study on degradation stability and optoelectronic performance of the FA-based phase
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22 pure 2D-MHPs is still lacking. MHPs in single crystal form would be ideal for analyzing
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24 fundamental properties due to large grain boundaries, lesser defects, long diffusion length, and
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26 sustainability against high current densities.^[26-29] Therefore, the growth of homogeneous large-
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28 area 2D MHP single crystals is pivotal for the stability analysis to explore them into a wide range
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30 of optoelectronic applications. However, synthesis and large size single crystal growth of phase
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32 pure 2D-MHPs, particularly for higher 'n' series, is quite challenging due to extensive organic
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34 moieties, quasi/multiphase contamination, and random stacking due to rapid nucleation. Randomly
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36 stacked 2D hybrid perovskite single crystals with phenyl ethyl ammonium (PEA) cation grown
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38 from a supersaturated HI solution by the slow cooling method were found to exhibit ultralow self-
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40 doping concentrations.^[30]

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43 Recently, we have demonstrated the growth of phase pure, high-quality millimeter size
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45 homologous 2D-MHPs (BA)₂(MA)_{n-1}Pb_nX_{3n+1} (n=1,2, and 3) single crystals by slow evaporation
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47 at constant temperature (SECT) solution growth method for optical resonators.^[31,32] Herein, we
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49 used the SECT method to grow a high-quality centimeter long (BA)₂FAPb₂I₇ single crystal to
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4 systematically investigate degradation stability and optoelectronic properties. We found that the
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6 large area FA-based 2D hybrid perovskite single crystal shows good stability against atmospheric
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8 degradation and laser irradiation compared to their potential counterpart of the MA-based
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10 (BA)₂MAPb₂I₇ single crystal. Due to the enhanced stability, the (BA)₂FAPb₂I₇ single-crystal
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12 exhibits durable photodetector performance on both the rigid and the flexible polymer substrates.
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15 16 **2. Results and Discussion**

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18 The phase pure 2D (BA)₂(FA)Pb₂I₇ compound having two perovskite layer thicknesses
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20 (n=2) was synthesized by the stoichiometric reaction of PbI₂, HC(=NH)NH₃I (FAI), and C₄H₉NH₃I
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22 (BAI) in I₂ free HI solution. H₃PO₂ was used to form an I₂-free HI solution. More details about the
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24 synthesis are given in the experimental section. An accurate concentration of BA is crucial in this
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26 reaction because it effectively controls reaction kinetics. The concentration of each reactant was
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28 optimized based on the earlier report by Stoumpos *et al.* [10]. The rectangular dark brown crystalline
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30 flakes obtained from the above reaction were used to grow high-quality single crystals. The
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32 schematic illustration of **Figure 1a** reveals the SECT growth process. Accordingly, a saturated
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34 solution with solvent-solute equilibrium was achieved by dissolving as-synthesized (BA)₂FAPb₂I₇
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36 flakes (Figure S1) into the I₂ free HI solvent at 80°C; it was then subjected to slow evaporation at
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38 a constant temperature. Due to solvent evaporation, the excessive solute nucleates as a tiny seed
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40 at the different positions of the beaker bottom by spontaneous nucleation. After three days of the
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42 growth period, well-faceted, rectangular-shaped crystals were obtained. The photo image of the
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44 as-grown centimeter-long 2D (BA)₂FAPb₂I₇ single crystal of dimension 1.1 cm x 0.4 cm is shown
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46 in Figure 1b.
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54 The phase purity and single crystallinity of the as-grown 2D (BA)₂FAPb₂I₇ perovskite
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56 single crystal was examined by X-ray diffraction (XRD) and transmission electron microscopic
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4 (TEM) analysis. Figure 1c shows the XRD pattern of 2D perovskite single-crystal indexed with
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6 respect to RP phase monoclinic crystal structure. The periodic repetitions of diffraction planes in
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8 the XRD pattern indicate single crystallinity. The characteristic features of the XRD pattern were
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10 typically used to precisely determine the number of perovskite layers (quantum well thickness) by
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12 means of the unit cell expansion in 2D perovskites. By introducing FA /or MA cation into the
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14 (BA)₂PbI₄ (n=1) crystal structure (where the thickness of one layer of perovskite is ~6.41 Å), the
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16 unit cell incrementally expands by the addition of each perovskite layer at a time. In the case of
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18 RP phase 2D hybrid perovskite n=2, the thickness of the perovskite increased to two layers,
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20 ~12.55 Å (Figure S2b). An addition of each perovskite layer expands the unit cell resulting in
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22 additional low angle reflection below $2\theta = 14^\circ$ in the XRD pattern. As shown in Figure 1c, the
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24 formation of two low angle diffraction peaks below $2\theta = 14^\circ$ in the XRD pattern reveals two
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26 perovskite layer thicknesses (n=2) in the crystal structure. Based on the XRD study, the
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28 predominant crystal growth orientation of the (BA)₂FAPb₂I₇ single crystal was found along the b-
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30 axis. The XRD results obtained from this study are good in agreement with the standard literature
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32 report.^[10, 33] The 2θ cut-off numbering ($2\theta = 14^\circ$) is based on the fact that, at this angle, the *d*
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34 spacing values matches the distance between the discrete perovskite layers in both the 2D [(111)
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36 reflection $d_{111}=6.26$ Å] and 3D [(110) reflection; $d_{110}=6.26$ Å] perovskites.^[10, 34] This numerical
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38 estimation of the crystal structure is highly recommended to study the (BA)₂(MA)_{n-1}Pb_nI_{3n+1} and
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40 related RP hybrid perovskite compounds.^[10] For 2D hybrid perovskite with n=2, the XRD shows
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42 two evenly spaced reflections (Figure S2), and other high member series, such as n=3, 4, and 5,
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44 shows a corresponding number of reflections below this cut-off value $2\theta = 14^\circ$ in the XRD patterns,
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46 which in turn confirm the number of perovskite layer and phase purity.^[10,35,36]
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4 The selective area diffraction (SAED) pattern (Figure 1d) obtained from TEM analysis
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6 reveals the linearly ordered bright diffraction spots, which further confirms single crystallinity.
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8 Moreover, the high quality of the large area single crystal (TEM image: Figure S3, Supporting
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10 Information) allows us to record lattices with hexagonal patterns attributed to the conductive
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12 perovskite top layer by using a high-resolution transmission electron microscopic (HR-TEM)
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14 study (Figure 1e). The formation of a conductive perovskite top surface would be beneficial for
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16 electronic applications. Figure 1f shows the crystal structure of $(\text{BA})_2(\text{FA})\text{Pb}_2\text{I}_7$, where PbI_6
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18 perovskites are oriented along the crystal surface. The thickness of one-layer perovskite was
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20 estimated to be 0.6 nm in hybrid 2D RP-phase architectures.
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26 The optical homogeneity and phase purity of the 2D perovskite single crystal was further
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28 examined by spatially correlated photoluminescence (PL) spectral studies. **Figure 2a** shows the
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30 optical microscopic image of a top surface of the $(\text{BA})_2\text{FAPb}_2\text{I}_7$ ($n=2$) single crystal with the
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32 dimension of about $5\ \mu\text{m} \times 5\ \mu\text{m}$, respectively, in which the surface morphology reveals a pyramid-
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34 like growth pattern as observed in MA-based RP phase 2D-MHPs.^[31] Figure 2b depicts the optical
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36 absorption spectrum obtained with strong narrowband absorption at the optical band gap energy
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38 of 2.17 eV, indicating the direct bandgap semiconducting behavior. Figure 2c shows the PL
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40 emission spectra measured at the excitation wavelength of 488 nm. An intense PL occurred at the
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42 wavelength of emission maxima centered at 577 nm having full-width half maxima (FWHM) ~ 23
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44 nm.^[37] There are no additional PL signals related to the quasi-phase contamination reveals the
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46 phase purity of the single crystal. The PL spectral mapping was carried out over the large area of
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48 the grown crystal to examine the PL homogeneity (Figure 2d). The spatially resolved PL spectral
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50 image was obtained with a collection of total mapping pixels of 64×64 is shown in Figure 2e. The
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52 uniform spectral distribution of emission wavelength over the entire crystal region reveals the
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4 optical homogeneity of the as-grown single crystal. The histogram (Figure 2f) reveals the
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7 581 nm, indicating that the superior uniformity of the as-grown 2D perovskite crystal.
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11 To investigate the material stability, the optical properties and surface features of the 2D
12 (BA)₂FAPb₂I₇ single crystal exposed under open laboratory atmospheric conditions (relative
13 humidity ~68% and room-temperature ~29°C) were monitored using PL spectroscopy and optical
14 microscopy (OM) at regular time intervals, 0, 30, and 60 min, respectively. The results were
15 compared with (BA)₂MAPb₂I₇ single crystal counterparts under the same conditions to evaluate
16 the relative enhancement instability. Both single crystals were exfoliated using the standard scotch
17 tape procedure on a silicon wafer. As shown in **Figure 3a**, the (BA)₂FAPb₂I₇ crystal exhibits stable
18 PL emission and no atmospheric degradation even after 60 min at open atmospheric exposure. At
19 the same time, the (BA)₂MAPb₂I₇ loss PL emission intensity half of its original value within 30
20 min and completely vanishes in 60 min time duration (Figure 3b).^[33,38] This phenomenon was
21 verified with the aid of OM analysis. From the OM images, the (BA)₂MAPb₂I₇ single-crystal starts
22 to degrade rapidly when the time is prolonged (Figure S4, Supporting Information). 50% of the
23 total surface degrades in 30 min of time duration and complete degradation after 60 min, which
24 leads to a total loss of PL emission (Figure 3b). On the other hand, the (BA)₂FAPb₂I₇ single crystal
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49 The stability of these 2D hybrid perovskites under atmospheric exposure can be correlated to
50 atmospheric moisture absorption. The moisture in the air could lead to material degradation due to
51 water absorption by organic moieties in the hybrid perovskites. However, in the 2D Ruddeldsen-
52 Popper (RP) phase (BA)₂(FA)Pb₂I₇ (n=2) and (BA)₂(MA)Pb₂I₇ (n=2) hybrid perovskites, the long-
53 chain butylammonium (BA) cation is a hydrophobic spacer, which is reluctant to moisture
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4 absorption. It has been widely proved that the long-chain BA spacer is crucial for enhanced
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6 environment stability in the 2D layered structure perovskites. Tsai *et al.* reported 2D RP phase
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8 $(\text{BA})_2(\text{MA})_3\text{Pb}_4\text{I}_{14}$ ($n=4$) perovskite with superior long-term stability with slower degradation
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10 under humid atmosphere as compared to the 3D MAPbI_3 , and they claimed it enhanced the stability
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12 of the $(\text{BA})_2(\text{MA})_3\text{Pb}_4\text{I}_{14}$ due to the hydrophobicity of the BA spacer.^[39] Wang *et al.* demonstrated
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14 ambient–air-stable solar cells formed by BA-incorporated mixed-cation (FA/Cs) perovskite. The
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16 hybrid perovskite with BA/FA/Cs showed long-term stability compared to the FA/Cs alone
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18 perovskite.^[40] In our study, atmospheric degradation in $(\text{BA})_2(\text{FA})\text{Pb}_2\text{I}_7$ ($n=2$) and
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20 $(\text{BA})_2(\text{MA})\text{Pb}_2\text{I}_7$ ($n=2$) might indicate the role of small methylammonium (MA) and
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22 formamidinium (FA) cations. As an experimental fact, the $(\text{BA})_2(\text{MA})\text{Pb}_2\text{I}_7$ ($n=2$) degrades faster
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24 than $(\text{BA})_2(\text{FA})\text{Pb}_2\text{I}_7$ ($n=2$). The rapid degradation of $(\text{BA})_2(\text{MA})\text{Pb}_2\text{I}_7$ indicates moisture-assisted
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26 internal migration of loosely bound MA cations due to a small ionic radius and low tolerance in
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28 the perovskite lattice.^[21,22] On the other hand, the large ionic radius with the tolerance factor of
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30 FA cation leads to a stable perovskite crystal structure with enhanced atmospheric stability under
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32 the protection of BA cation.^[23]

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41 To distinguish between $(\text{BA})_2\text{FAPb}_2\text{I}_7$ and $(\text{BA})_2\text{MAPb}_2\text{I}_7$ single crystals, the presence of
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43 different functional groups was examined by Fourier Transform infrared spectroscopy (FT-IR)
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45 spectroscopy.^[41,42] Figure 4a shows that the FT-IR spectra were measured in the wavenumber
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47 ranges from 400 to 4000 cm^{-1} . The formation of IR bands between 3100 and 3420 cm^{-1} is assigned
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49 for the N-H stretching vibration from the primary alkyl ammonium cation. The bands observed
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51 between 2800 cm^{-1} and 2950 cm^{-1} correspond to alkyl C-H stretching. The sharp peaks formed at
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53 1573 cm^{-1} and 1463 cm^{-1} for the $(\text{BA})_2\text{MAPb}_2\text{I}_7$ are assigned for CH_3 bending vibration. The
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55 formation of a sharp band at a wavenumber 906 cm^{-1} was observed for the $(\text{BA})_2\text{MAPb}_2\text{I}_7$ assigned
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4 for the characteristic rocking vibration of $-\text{CH}_3\text{NH}_3$, which is a complete absence for the
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7 $(\text{BA})_2\text{FAPb}_2\text{I}_7$ (Figure 4b) single crystal. On the other hand, the characteristic $-\text{C}=\text{N}$ stretching
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9 vibration at 1710 cm^{-1} is only observed for the $(\text{BA})_2\text{FAPb}_2\text{I}_7$ compound (Figure 4c).
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12 By considering the advantage of all benefits, such as good crystallinity with a large area,
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14 strong optical absorption, and chemical/optical degradation stability, the phase pure 2D
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16 $(\text{BA})_2\text{FAPb}_2\text{I}_7$ single crystal was used as photoactive material on both rigid and flexible
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18 optoelectronic devices. A few-layered $(\text{BA})_2\text{FAPb}_2\text{I}_7$ - field-effect transistor (FET) was fabricated
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20 by using a standard shadow mask technique. The $(\text{BA})_2\text{FAPb}_2\text{I}_7$ nanosheets were obtained by the
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22 exfoliation scotch tape process and transferred onto the silicon substrate on the top with 300 nm
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24 thick SiO_2 dielectric layers (Figure S5, Supporting Information). Then, the Cr/Au metal electrodes
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26 of thicknesses 5 nm/70 nm were deposited at a high vacuum using a thermal evaporation system.
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31 **Figure 5a** represents the schematic illustration of $(\text{BA})_2\text{FAPb}_2\text{I}_7$ -FET under 488 nm laser
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33 irradiation. The channel (perovskite) current between drain-source (I_{ds}) under bias voltage (V_{ds})
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35 was measured, the plot of $I_{\text{ds}}-V_{\text{ds}}$ shown in Figure 5b and (Figure S6a, Supporting Information for
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37 $(\text{BA})_2\text{MAPb}_2\text{I}_7$) in the presence and absence of light. Upon increasing illumination light intensity,
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39 the channel current was found to increase linearly. Figure 5c displays the plot of generated
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41 photocurrent (I_{ph}) as a function of the incident laser power (P), where $I_{\text{ph}} = I_{\text{light}} - I_{\text{dark}}$ was calculated
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43 by subtracting the I_{ds} obtained in the dark (represented by I_{dark}) from that under illumination
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45 (denoted by I_{light}) at $V_{\text{ds}} = 10\text{ V}$, and $V_{\text{g}} = 0\text{ V}$ (Figure S6b, Supporting Information for
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47 $(\text{BA})_2\text{MAPb}_2\text{I}_7$). Photoresponsivity (R_{λ}) is a significant factor to critic the sensitivity of a
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49 photodetector under the light stimulus, which is defined as the photocurrent produced per unit
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51 power of the incident light on active channel area, is expressed as $R_{\lambda} = \Delta I_{\lambda} / (P_{\lambda} S)$, where ΔI_{λ} is the
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53 generated photocurrent, P_{λ} is the incident light power, S is the illuminated area. Figure 5d
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4 demonstrates the R_λ for incident laser intensity (P) values at 488 nm. The maximum R_λ value of
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6 $\sim 5.0 \text{ AW}^{-1}$ at $3.5 \mu\text{W}$ ($V_{\text{ds}} = 10 \text{ V}$ and $V_{\text{g}} = 0 \text{ V}$) was obtained from $(\text{BA})_2\text{FAPb}_2\text{I}_7$ -FET. The R_λ
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8 value of $(\text{BA})_2\text{FAPb}_2\text{I}_7$ -FET is 25 times higher than that of the $(\text{BA})_2\text{MAPb}_2\text{I}_7$ -FET (0.2 mAW^{-1}),
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10 shown in Figure S6c. Besides, the obtained responsivity of $(\text{BA})_2\text{FAPb}_2\text{I}_7$ compound is compared
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12 with reported photodetectors fabricated using other 2D and 3D hybrid perovskites (**Table 1**
13
14 Supporting Information).^[43-50] There are few reports on the enhanced responsivity of 2D hybrid
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16 perovskites using different strategies, Tan *et al.* used graphene as a protection layer and top contact
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18 for effective carrier collection in their $(\text{BA})_2\text{PbBr}_4$ photodetector to enhance the stability and
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20 responsivity.^[51] Loi *et al.* demonstrated high photoresponsivity by hot-casting 2D/3D vertical
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22 hetero-junction perovskite films.^[52] In this study, we showed the photodetector response of the
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24 pristine $(\text{BA})_2\text{FAPb}_2\text{I}_7$ -FET without any additional strategies. Intriguingly, the R_λ of
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26 $(\text{BA})_2\text{FAPb}_2\text{I}_7$ -FET increases to 63 AW^{-1} by tuning V_{bg} from 0 to +120 V. The plot of transfer
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28 curve (I_{ds} vs. V_{bg}) under various V_{bg} from -80 to +120 V, as shown in Figure 5e. A solid increase
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30 in current (I_{ds}) with increasing (V_{bg}) was observed, which clearly reveals n-type semiconducting
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32 behavior, and confirms that the majority of charge carriers are electrons in the RP phase 2D
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34 $(\text{BA})_2\text{FAPb}_2\text{I}_7$ perovskite single crystal. The n-type semiconducting behavior is consistent with
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36 the recently reported $(\text{BA})_2\text{MAPb}_2\text{I}_7$ single-crystalline FET device.^[32] The channel current was
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38 observed to increase dramatically relative to the dark state: the R_λ increases by simply tuning the
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40 gate voltage (V_{bg}) as shown in Figure 5f. Remarkably, the R_λ increased significantly from 3.8 to
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42 63 AW^{-1} when V_{bg} was swept from 0 to +120 V. The obtained immense R_λ value can be attributed
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44 to the high-quality large-area perovskite crystal with lesser defects, stabilized crystal structure, and
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46 high absorption coefficient, along with the direct bandgap nature of $(\text{BA})_2\text{FAPb}_2\text{I}_7$ crystal. All the
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48 above features of FA-based 2D perovskites enable strong light absorption, efficient photocarrier
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4 generation, and the excited charge carrier's movement to the electrodes before it recombines at the
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6 channel during the light stimulus.
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10 In addition to other important photodetector parameters, such as specific detectivity (D^*) and
11 normalized gain, were analyzed. The D^* value help to identify the weak optical signals of a
12 photodetector is defined by $D^* = (S \cdot \Delta f)^{1/2} / \text{NEP}$, where S and Δf are the effective area and electrical
13 bandwidth of the photodetector, respectively, and NEP represents noise equivalent power. The
14 NEP is a pointer to the minimum optical power that a photodetector can use to discriminate signals
15 from noise. At low NEP value, the above equation is used simply as $D^* = R_\lambda S^{1/2} / (2eI_{\text{dark}})^{1/2}$, where
16 R_λ , S , e , and I_{dark} are the photoresponsivity, effective area, elementary charge, and dark current,
17 respectively.^[53] The obtained D^* value for the few-layered $(\text{BA})_2\text{FAPb}_2\text{I}_7$ is $\sim 3.5 \times 10^{11}$ Jones (at
18 $V_g = 0$ and $V_{\text{ds}} = 10$ V), shown in Figure S7a of Supporting Information. The acquired D^* value
19 for $(\text{BA})_2\text{FAPb}_2\text{I}_7$ single-crystal \sim three orders higher than the $(\text{BA})_2\text{MAPb}_2\text{I}_7$ and \sim two times
20 larger than the recently reported 2D-MHPs.^[54,55] Also, very close to the commercial diodes.^[56] The
21 normalized gain was calculated by the given relation $\Gamma_n = (E/e) (I_{\text{ph}}/p) (l^2/V)$, where E is the photon
22 energy, e is the elementary charge, V is the bias voltage, L is the channel length of the device.^[57]
23 From the above relation, the calculated value of Γ_n is $1.2 \times 10^{-6} \text{ V}^{-1} \text{ cm}^2$ (Figure S7b of Supporting
24 Information), which is two orders higher compared with $(\text{BA})_2\text{MAPb}_2\text{I}_7$ -FET (Figure S8 of
25 Supporting Information)
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29 The time-resolved ($I_{\text{ds}}-t$) measurements were conducted to show the photoresponse,
30 photoswitching stability, and durability of the $(\text{BA})_2\text{FAPb}_2\text{I}_7$ photodetector at room temperature.
31 The $I_{\text{ds}}-t$ plot was recorded by exciting the multi-layered $(\text{BA})_2\text{FAPb}_2\text{I}_7$ device with a train of ON-
32 OFF illumination at 488 nm. Figure S9a and S9b of supporting Information shows the
33 $(\text{BA})_2\text{FAPb}_2\text{I}_7/(\text{BA})_2\text{MAPb}_2\text{I}_7$ -FET device in response to an ON-OFF train cycle at $V_{\text{ds}} = 10$ V
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4 and $V_g = 0$ V, which exhibits a fast rise of current (I_{ds}) under illumination and a quick drop followed
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7 by a much slower relaxation after switching off the laser. From the curve $I_{ds}-t$, the rising time of
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9 the photo-generated signal was found to be ~ 20 ms. The falling time contains two components
10 with a fast decay of ~ 20 ms (corresponding to the earlier 50% decrease) and slow relaxation of 1
11 s (in the later half decline). In response to consecutive photoswitching, the robustness and stability
12 of our $(BA)_2FAPb_2I_7$ single-crystal photo device was examined by a train of pulsed illumination
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14 over a long period (~ 1 h); the device shows good stability under ambient condition, as
15 demonstrated by the $I_{ds}-t$ measurements presented in the **Figure 6a**. The photoswitching
16 measurements were conducted multiple times to demonstrate the reproducibility of long-term
17 stability in $(BA)_2FAPb_2I_7$ (Figure S10). The $I_{ds}-t$ plot of $(BA)_2MAPb_2I_7$ shown in Figure 6b, where
18 the on-off curve starts to degrade after 30 minutes and wholly lost signal after 60 min, which is
19 well consisted with results of optical degradation shown in Figure 3b.
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34 We have further studied the 2D $(BA)_2FAPb_2I_7$ crystal performance on a flexible photodetector
35 using polyethylene terephthalate (PET) substrate. The $(BA)_2FAPb_2I_7$ photodetector was fabricated
36 on a PET polymer substrate by the scotch tape exfoliation procedure and followed by the Au
37 electrode deposition as schematically illustrated in **Figure 7a**. As observed in $(BA)_2FAPb_2I_7$
38 photodetector on a rigid substrate, flexible photodetector also clutches a good contact, clearly
39 noticed in the $I_{ds}-V_{ds}$ measurements (Figure 7b). The device channel is more conductive with a
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41 488 nm wavelength laser illumination because of numerous photoexcited carriers in the
42 $(BA)_2FAPb_2I_7$ perovskite. The photocurrent was measured at laser illumination with different
43 power intensities to evaluate the performance of the $(BA)_2FAPb_2I_7$ single crystal at bending and
44 non-bending states. The photocurrent rises with increasing laser intensity, from which the photo
45 responsivity was calculated to be 0.1 A/W, at 0.2 mW and 10 V, as shown in Figure 7c. Figure 7d
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4 shows the responsivity under different bending curvatures (1 cm and 2 cm). We witnessed that the
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6 responsivity value changes and reduces at bending states compared with the balanced state. The
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8 fall of responsivity in a bending state might be attributed to the induced strain in the channel, thus
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10 reducing electron-hole pair generation efficiency.^[58] In addition, the formation of the crack on the
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12 Au metal pads during the bending further reduces the responsivity values.^[59] Typically, the flaws
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14 in the metal electrode build high contact resistance with (BA)₂FAPb₂I₇ perovskite, which
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16 additionally reduces the responsivity. Overall, the device performance showed that 2D
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18 (BA)₂FAPb₂I₇ can effectively work under strain-induced modulation without much deterioration.
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20 These outcomes disclose a favorable flexible nature of (BA)₂FAPb₂I₇ perovskite semiconductors
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22 for excellent and workable practical applications.
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29 **3. Conclusion**

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32 In summary, we have successfully demonstrated a stable and durable photodetector
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34 fabricated using formamidinium based phase-pure centimeter-long hybrid 2D (BA)₂FAPb₂I₇
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36 perovskite single crystals having good crystallinity and spectral uniformity. The (BA)₂FAPb₂I₇
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38 single crystal showed a tremendous enhancement in the chemical and optical degradation stability
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40 against atmospheric oxidation and continuous laser irradiation. The robustness of the stable 2D
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42 perovskite material shows high photoresponsivity and specific detectivity on both the rigid and
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44 flexible ultraviolet photodetectors, which are much higher than that of the MA-based 2D/3D
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46 counterparts and comparable to the commercialized Si photodetector. From this study, the FA-
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48 based 2D hybrid perovskite single-crystal is highly promising for future visible detectors and
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50 wearable electronics. Moreover, replacing iodide with other halides, such as chloride and bromide,
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52 would be beneficial by adding degrees of freedom in tunability for a wide range of stable and low-
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54 cost optoelectronic applications.
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4. Experimental Section

Synthesis

The raw materials, PbO, 57% aqueous HI, H₃PO₂, C₄H₉NH₂ (BA), CH₃NH₃Cl (MACl), and HC(NH₂)₂Cl (FACl) with purity (99.99%) were obtained from Aldrich chemicals. PbO (10 mmol) was added into the solution mixture of HI (76 mmol)/H₃PO₂ (15.5 mmol) at 70°C under constant magnetic stirring to form a dissolved bright yellow PbI₂ solution. Concurrently, the C₄H₉NH₃I (BAI) was prepared by neutralizing BA with HI by continuous magnetic stirring for 2 hr in an ice bath. The measured quantity of FACl (5 mmol) was dissolved into the hot PbI₂ solution, which initially led to the precipitation of FAPbI₃ that was re-dissolved by heating with slightly excessive solvent to form a stable, clear, bright yellow solution. Finally, a neutralized BAI (7 mmol for n =2) solution was added dropwise into the clear, bright yellow solution of FAPbI₃ to form RP phase 2D (BA)₂FAPb₂I₇. The solution was left to cool at room temperature until dark red flakes were deposited. The deposited product was filtered and dried using suction filtration under a vacuum (Figure S1). For the synthesis of (BA)₂MAPb₂I₇, the same procedure was followed with MACl instead of FACl. Accordingly, the MACl (5 mmol) was added into the PbI₂ solution to form the MAPI₃ bright yellow solution. To the clear yellow bright solution of MAPI₃, the BAI (7 mmol) solution was added to form the (BA)₂MAPb₂I₇ compound.

Single Crystals Growth

The synthesized (BA)₂FAPb₂I₇ flakes were used to grow centimeter-long single crystals using the HI/H₃PO₂ solvent medium by the slow evaporation at constant temperature (SECT) solution-growth method.^[31] The HI/H₃PO₂ solution was saturated by adding 2D perovskite with magnetic stirring at a constant temperature of 80°C in an oil bath. The beakers containing a

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4 saturated bright yellow solution were tightly covered with a paraffin sheet. The saturated solution
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6 was then allowed for controlled evaporation by making holes at the top of the paraffin sheet.
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8 Initially, small nuclei formed at different positions at the bottom of the beaker. Then, they started
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10 to grow into large-sized single crystals from a supersaturated solution. After three days, the crystals
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12 at the bottom of the beakers become thick individual platelets of centimeter-sized homologous 2D
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14 RP-phase perovskite single crystals. The grown crystals were then gently harvested and dried at
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18 55 °C in an oven under reduced pressure to remove the surface adsorbed solvent.
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21 **Materials Characterization**

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25 The crystal structure and phase purity of the (BA)₂FAPb₂I₇ single crystal were studied by
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27 X-ray diffraction analysis with an X'Pert PRO-PANalytical X-ray diffractometer using CuK α (λ
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29 = 1.5406 Å) source, step size 0.01°, scan speed 0.5 s/step and operation voltage 40 kV. The
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31 microstructural features of the grown crystal were examined by Transmission Electron Microscopy
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33 (TEM) using JEOL, JEM-2100F operated at 200 kV. A home-built setup was used for the
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35 photoluminescence spectral mapping with a 450 nm pulse laser (Pico Quant) as an excitation light
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37 source. The optical absorption spectra were recorded using Jacobs V-670 UV-Vis spectrometer.
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39 The Fourier Transform infrared spectroscopy (FTIR) was analyzed by Perkin Elmer model
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46 spectrum in the transmittance mode in the wavenumber range 400-4000 cm⁻¹.

47 **Mechanical Exfoliation-Few layered**

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51 As-grown 2D (BA)₂FAPb₂I₇ and (BA)₂MAPb₂I₇ perovskite single crystals were used to
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53 obtain few-layered nanosheets using a scotch tape mechanical exfoliation technique. Briefly, the
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55 bulk 2D perovskite single crystals were gently rubbed and sliced multiple times on the top of the
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57 scotch tape to obtain a few-layered nanosheet and transferred to the Si wafer containing a 300 nm
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4 thick SiO₂ dielectric layer and PET substrates. The transferred perovskite layers were examined
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6 with an optical microscope (Olympus, BX 51M) equipped with a charge-coupled device (CCD)
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8 camera (Leica, DFC495).
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10 11 **FET-device fabrication and Optoelectronic Properties**

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15 The TEM copper grid was used to pattern the electrode on the few-layered (BA)₂FAPb₂I₇
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17 and (BA)₂MAPb₂I₇ nanosheets with the aid of a micromanipulator. Also, the TEM grid was used
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19 as a shadow mask for the deposition of source and drain electrodes of Cr (5 nm)/Au (70 nm) using
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21 a thermal evaporation unit. The optoelectronic properties of the (BA)₂FAPb₂I₇ and
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23 (BA)₂MAPb₂I₇-FETs were studied using a probe station (Lakeshore, TTPX) equipped with a
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25 source electrometer (Keithley, 2636A) and an optical system, including a He–Ne laser (JDS
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27 Uniphase, Novette 1507), a power meter (Ophir, Nova II), an optical beam shutter (Thorlabs, SH1).
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32 **Conflicts of interest**

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35 There are no conflicts of interest to declare.
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References

1. J. Y. Kim, J.-W. Lee, H. S. Jung, H. J. Shin, N.-G. Park, *Chem. Rev.* **2020**, *120*, 7867.
2. L. Chouhan, S. Ghimire, C. Subrahmanyam, T. Miyasaka, V. Biju, *Chem. Soc. Rev.* **2020**, *49*, 2869.
3. S. D. Stranks, H. J. Snaith, *Nat. Nanotechnol.* **2015**, *10*, 391.
4. A. Miyata, A. Mitioglu, P. Plochocka, O. Portugall, J. T.-W. Wang, S. D. Stranks, H. J. Snaith, R. J. Nicholas, *Nat. Phys.* **2015**, *11*, 582.
5. Q. Dong, Y. Fang, Y. Shao, P. Mulligan, J. Qiu, L. Cao, J. Huang, *Science* **2015**, *347*, 967.
6. L. Dou, Y. Yang, J. You, Z. Hong, W.-H. Chang, G. Li, Y. Yang, *Nat. Commun.* **2014**, *5*, 5404.
7. M. A. Green, A. H.-Baillie, H. J. Snaith, *Nat. Photon.* **2014**, *8*, 506.
8. Y.-H. Kim, H. Cho, J. H. Heo, T.-S. Kim, N. S. Myoung, C.-L. Lee, S. H. Im, T.-W. Lee, *Adv. Mater.* **2015**, *27*, 1248.
9. P. Liu, X. He, J. Ren, Q. Liao, J. Yao, H. Fu, *ACS Nano* **2017**, *11*, 5766.
10. C. C. Stoumpos, D. H. Cao, D. J. Clark, J. Young, J. M. Rondinelli, J. I. Jang, J. T. Hupp, M. G. Kanatzidis, *Chem. Mater.* **2016**, *28*, 2852.
11. H. Tsai, W. Nie, J.-C. Blancon, C. C. Stoumpos, R. Asadpour, B. Harutyunyan, A. J. Neukirch, R. Verduzco, J. J. Crochet, S. Tretiak, L. Pedesseau, J. Even, M. A. Alam, G. Gupta, J. Lou, P. M. Ajayan, M. J. Bedzyk, M. G. Kanatzidis, A. D. Mohite, *Nature*, **2016**, *536*, 312.
12. I. C. Smith, E. T. Hoke, D. S.-Ibarra, M. D. McGehee, H. I. Karunadasa, *Angew. Chem.* **2014**, *126*, 11414.
13. D. H. Cao, C. C. Stoumpos, O. K. Farha, J. T. Hupp, M. G. Kanatzidis, *J. Am. Chem. Soc.* **2015**, *137*, 7843.
14. H. Lai, D. Lu, Z. Xu, N. Zheng, Z. Xie, Y. Liu, *Adv. Mater.* **2020**, *32*, 2001470.
15. H.-P. Wang, S. Li, X. Liu, Z. Shi, X. Fang, J. H. He, *Adv. Mater.* **2021**, *33*, 2003309.
16. L. Zhang, C. Sun, T. He, Y. Jiang, J. Wei, Y. Huang, M. Yuan, *Light Sci. Appl.* **2021**, *10*:61.
17. Z. Ren, J. Yu, Z. Qin, J. Wang, J. Sun, C. C. S. Chan, S. Ding, K. Wang, R. Chen, K. S. Wong, X. Lu, W.-J. Yin, W. C. H. Choy, *Adv. Mater.* **2021**, *33*, 2005570.
18. C. Qin, A. S. D. Sandanayaka, C. Zhao, T. Matsushima, D. Zhang, T. Fujihara, C. Adachi, *Nature* **2020**, *585*, 53.
19. Z. Liu, M. Hu, J. Du, T. Shi, Z. Wang, Z. Zhang, Z. Hu, Z. Zhan, K. Chen, W. Liu, J. Tang, H. Zhang, Y. Leng, R. Li, *ACS Nano* **2021**, *15*, 6900.
20. F. Liu, L. Wang, J. Wang, F. Wang, Y. Chen, S. Zhang, H. Sun, J. Liu, G. Wang, Y. Hu, C. Jiang, *Adv. Funct. Mater.* **2021**, *31*, 2005662.
21. B. Conings, J. Drijkoningen, N. Gauquelin, A. Babayigit, J. D. Haen, L. D. Olieslaeger, A. Ethirajan, J. Verbeeck, J. Manca, E. Mosconi, F. D. Angelis, H.-G. Boyen, *Adv. Energy Mater.* **2015**, *5*, 1500477.
22. A. Senocrate, G. Y. Kim, M. Gratzel, J. Maier, *ACS Energy Lett.* **2019**, *4*, 2859.
23. Q. Han, S.-H. Bae, P. Sun, Y.-T. Hsieh, Y. Yang, Y. S. Rim, H. Zhao, Q. Chen, W. Shi, G. Li, Y. Yang, *Adv. Mater.* **2016**, *28*, 2253.

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24. J. Jeong, M. Kim, J. Seo, H. Lu, P. Ahlawat, A. Mishra, Y. Yang, M. A. Hope, F. T. Eickemeyer, M. Kim, Y. J. Yoon, I. W. Choi, B. P. Darwich, S. J. Choi, Y. Jo, J. H. Lee, B. Walker, S. M. Zakeeruddin, L. Emsley, U. Rothlisberger, A. Hagfeldt, D. S. Kim, M. Gratzel, J. Y. Kim, *Nature* **2021**, 592, 381.
 25. Y. Chen, M. He, J. Peng, Y. Sun, Z. Liang, *Adv. Sci.* **2016**, 3, 1500392.
 26. T. Liu, J. Zhang, M. Qin, X. Wu, F. Li, X. Lu, Z. Zhu, A. K.-Y. Jen, *Adv. Funct. Mater.* **2021**, 31, 2009515.
 27. Y. Liu, H. Xiao, W. A. Goddard III, *Nano Lett.* **2016**, 16, 3335.
 28. E. Penzo, A. Loiudice, E. S. Barnard, N. J. Borys, M. J. Jurow, M. Lorenzon, I. Rajzbaum, E. K. Wong, Y. Liu, A. M. Schwartzberg, S. Cabrini, S. Whitelam, R. Buonsanti, A. W.-Bargioni, *ACS Nano* **2020**, 14, 6999.
 29. J. Wang, J. Li, S. Lan, C. Fang, H. Shen, Q. Xiong, D. Li, *ACS Nano* **2019**, 13, 5473.
 30. W. Peng, J. Yin, K.-T. Ho, O. Ouellette, M. D. Bastiani, B. Murali, O. E. Tall, C. Shen, X. Miao, J. Pan, E. Alarousu, J.-H. He, B. S. Ooi, O. F. Mohammed, E. Sargent, O. M. Bakr, *Nano Lett.* **2017**, 17, 4759.
 31. C. M. Raghavan, T.-P. Chen, S.-S. Li, W.-L. Chen, C.-Y. Lo, Y.-M. Liao, G. Haider, C.-C. Lin, C.-C. Chen, R. Sankar, Y.-M. Chang, F.-C. Chou, C.-W. Chen, *Nano Lett.* **2018**, 18, 3221.
 32. M.-K. Li, T.-P. Chen, Y.-F. Lin, C. M. Raghavan, W.-L. Chen, S.-H. Yang, R. Sankar, C.-W. Luo, Y.-M. Chang, C.-W. Chen, *Small*, **2018**, 14, 1803763.
 33. E. Shi, S. Deng, B. Yuan, Y. Gao, Akriti, L. Yuan, C. S. Davis, D. Zemlyanov, Y. Yu, L. Huang, L. Dou *ACS Nano* **2019**, 13, 1635.
 34. D. H. Cao, C. C. Stoumpos, O. K. Farha, J. T. Hupp, M. G. Kanatzidis, *J. Am. Chem. Soc.* **2015**, 137, 24, 7843.
 35. D. T. Gangadharan, D. Ma, *Energy Environ. Sci.*, **2019**, 12, 2860.
 36. J.-C. Blancon, H. Tsai, W. Nie, C. C. Stoumpos, L. Pedesseau, C. Katan, M. Kepenekian, C. M. M. Soe, K. Appavoo, M. Y. Sfeir, S. Tretiak, P. M. Ajayan, M. G. Kanatzidis, J. Even, J. J. Crochet, A. D. Mohite, *Science*, **2017**, 355, 1288.
 37. J. Yan, W. Fu, X. Zhang, J. Chen, W. Yang, W. Qiu, G. Wu, F. Liu, P. Heremans, H. Chen, *Mater. Chem. Front.* **2018**, 2, 121.
 38. F. Wang, J. Ma, F. Xie, L. Li, J. Chen, J. Fan, N. Zhao, *Adv. Funct. Mater.* **2016**, 26, 3417.
 39. H. Tsai, W. Nie, J.-C. Blancon, C. C. Stoumpos, R. Asadpour, B. Harutyunyan, A. J. Neukirch, R. Verduzco, J. J. Crochet, S. Tretiak, L. Pedesseau, J. Even, M. A. Alam, G. Gupta, J. Lou, P. M. Ajayan, M. J. Bedzyk, M. G. Kanatzidis, A. D. Mohite, *Nature*, **2016**, 536, 312.
 40. Z. Wang, Q. Lin, F. P. Chmiel, N. Sakai, L. M. Herz, H. J. Snaith, *Nat. Energy*, **2017**, 2, 17135.
 41. Z. Zhou, S. Pang, F. Ji, B. Zhang, G. Cui, *Chem. Commun.* **2016**, 52, 3828.
 42. K. H.-Kimball, Y. Nagaoka, C. Cao, E. Chaykovsky, O. Chen, *J. Mater. Chem. C*, **2017**, 5, 5680.
 43. R. Dong, C. Lan, X. Xu, X. Liang, X. Hu, D. Li, Z. Zhou, L. Shu, S. Yip, C. Li, S.-W. Tsang, J. C. Ho, *ACS Appl. Mater. Interfaces* **2018**, 10, 19019.
 44. J. Chen, Y. Wang, L. Gan, Y. He, H. Li, T. Zhai, *Angew. Chem. Int. Ed.* **2017**, 56, 14893.L.

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46
47
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55
56
57
58
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62
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65
45. L. Min, W. Tian, F. Cao, J. Guo, L. Li, *Adv. Mater.* **2021**, 2101714.S.
 46. R. Saraf, V. Maheshwari, *ACS Appl. Mater. Interfaces* **2018**, *10*, 21066.
 47. S. Ma, G. Jang, S. Kim, H.-C. Kwon, S. Goh, H. Ban, J. H. Cho, J. Moon, *ACS Appl. Mater. Interfaces* **2020**, *12*, 41674.
 48. Z. Chen, Z. Kang, C. Rao, Y. Cheng, N. Liu, Z. Zhang, L. Li, Y. Gao, *Adv. Electron. Mater.* **2019**, *5*, 1900168.
 49. Z.-X. Zhang, L.-H. Zeng, X.-W. Tong, Y. Gao, C. Xie, Y. H. Tsang, L.-B. Luo, Y.-C. Wu, *J. Phys. Chem. Lett.* **2018**, *9*, 1185.
 50. J. Li, S. Yuan, G. Tang, G. Li, D. Liu, J. Li, X. Hu, Y. Liu, J. Li, Z. Yang, S. F. Liu, Z. Liu, F. Gao, F. Yan, *ACS Appl. Mater. Interfaces* **2017**, *9*, 42779.
 51. Z. Tan, Y. Wu, H. Hong, J. Yin, J. Zhang, L. Lin, M. Wang, X. Sun, L. Sun, Y. Huang, K. Liu, Z. Liu, H. Peng, *J. Am. Chem. Soc.* **2016**, *138*, 16612.
 52. H.-L. Loi, J. Cao, X. Guo, C.-K. Liu, N. Wang, J. Song, G. Tang, Y. Zhu, F. Yan, *Adv. Sci.* **2020**, *7*, 2000776.
 53. R. K. Ulaganathan, R. Sankar, C.-Y. Lin, R. C. Murugesan, K. Tang, F.-C. Chou, *Adv. Electron. Mater.* **2020**, *6*, 1900794.
 54. L. Qian, Y. Sun, M. Sun, Z. Fang, L. Li, D. Xie, C. Li, L. Ding, *J. Mater. Chem. C* **2019**, *7*, 5353
 55. Z. Li, H. Li, K. Jiang, D. Ding, J. Li, C. Ma, S. Jiang, Y. Wang, T. D. Anthopoulos, Y. Shi, *ACS Appl. Mater. Interfaces* **2019**, *11*, 40204.
 56. X. Gong, M. Tong, Y. Xia, W. Cai, J. S. Moon, Y. Cao, G. Yu, C.-L. Shieh, B. Nilsson and A. J. Heeger, *Science* **2009**, *325*, 1665.
 57. C.-Y. Lin, R. K. Ulaganathan, R. Sankar, R. C. Murugesan, A. Subramanian, A. Rozhin, S. Firdoz, *J. Mater. Chem. C* DOI: 10.1039/d1tc02129j.
 58. H. J. Conley, B. Wang, J. I. Ziegler, R. F. Haglund, S. T. Pantelides, K. I. Bolotin, *Nano Lett.* **2013**, *13*, 3626.
 59. G.-H. Lee, Y.-J. Yu, X. Cui, N. Petrone, C.-H. Lee, M. S. Choi, D.-Y. Lee, C. Lee, W. J. Yoo, K. Watanabe, T. Taniguchi, C. Nuckolls, P. Kim, J. Hone, *ACS Nano* **2013**, *7*, 7931.

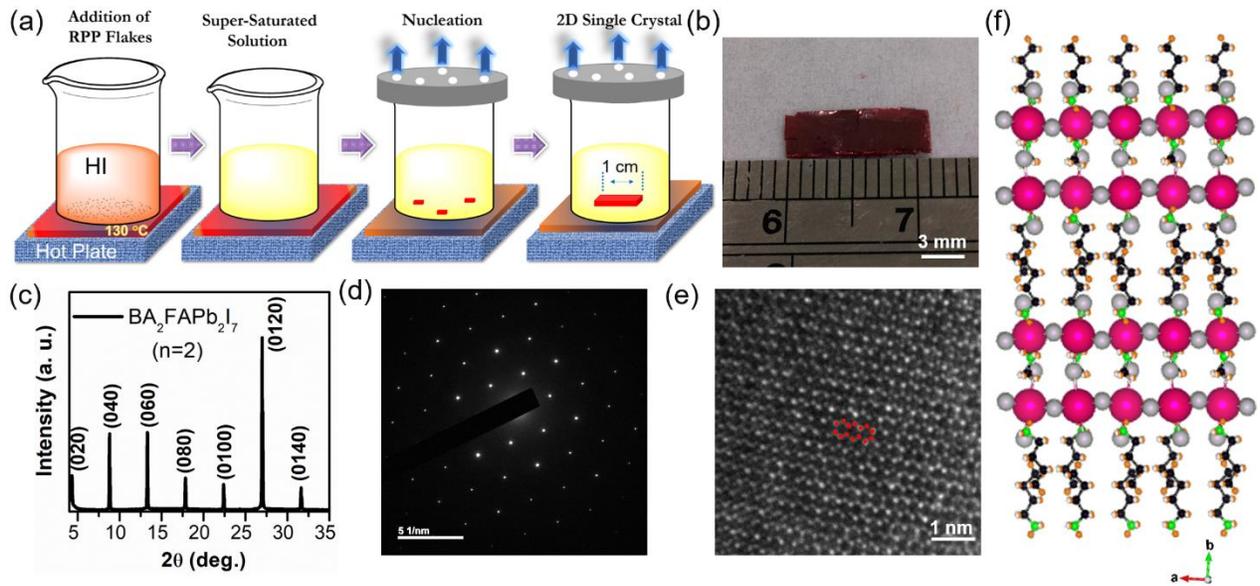


Figure 1. (a) Schematic illustration of the SECT method for growing high-member 2D perovskite single crystals. (b) Photograph of as-grown centimeter long 2D $(\text{BA})_2\text{FAPb}_2\text{I}_7$ perovskite single crystal by SECT method. (c) Indexed XRD pattern with repeated $(0k0)$ miller planes. (d) SAED pattern with linearly ordered bright diffraction spots confirms the single crystallinity. (e) A regular hexagonal arrangement of atoms was observed in the HR-TEM image, indicating a conductive perovskite layer on the top surface. (f) The RP phase 2D $(\text{BA})_2\text{FAPb}_2\text{I}_7$ crystal structure has two perovskite layers.

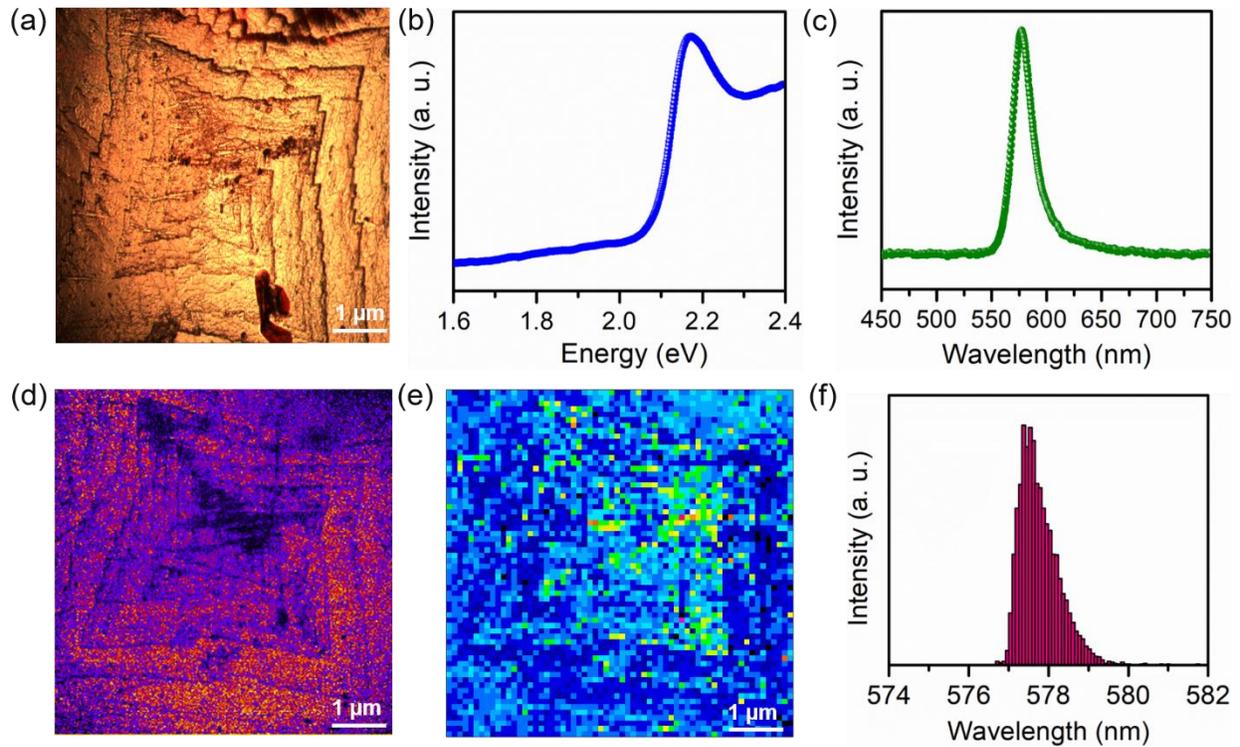


Figure 2. (a). Optical image of the $(\text{BA})_2\text{FAPb}_2\text{I}_7$ single crystal top surface. (b) The optical absorption spectrum reveals a sharp absorption band at the bandgap energy of 2.17 eV (c) Photoluminescence (PL) spectrum with intense emission centered at a wavelength of 577 nm (d) The crystal top surface used for the PL spectral mapping and (e) the spectral mapping image. (f) The histogram of the peak wavelength measured over the top surface and the emission spectra are distributed between 576 nm and 581 nm, confirming the uniformity of $(\text{BA})_2\text{FAPb}_2\text{I}_7$ single crystal.

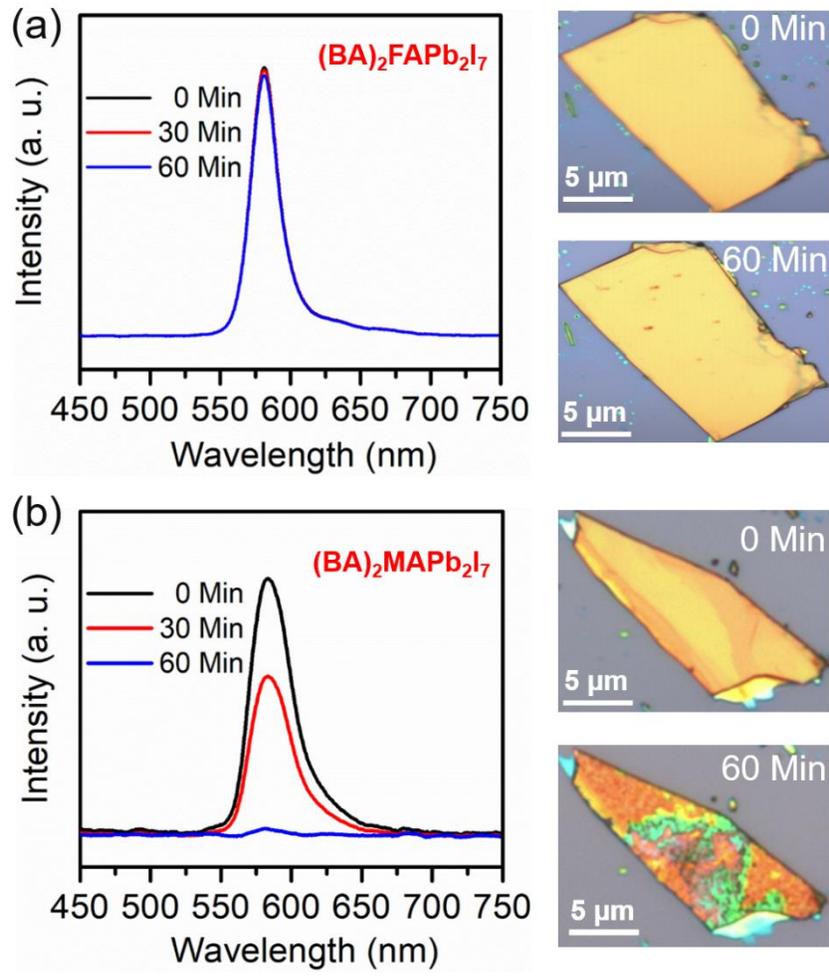


Figure 3. Optical stability of $(\text{BA})_2\text{FAPb}_2\text{I}_7$ and $(\text{BA})_2\text{MAPb}_2\text{I}_7$ single crystals: (a) The $(\text{BA})_2\text{FAPb}_2\text{I}_7$ crystal shows stable PL emission and no atmospheric degradation even after 60 min of exposure at the ambient environment (b) The $(\text{BA})_2\text{MAPb}_2\text{I}_7$, the PL emission intensity starts to diminish. Half of its intensity was lost within 30 minutes and completely vanishes in 60 minutes; this effect was clearly seen in the optical image of the right side, and the crystal degraded.

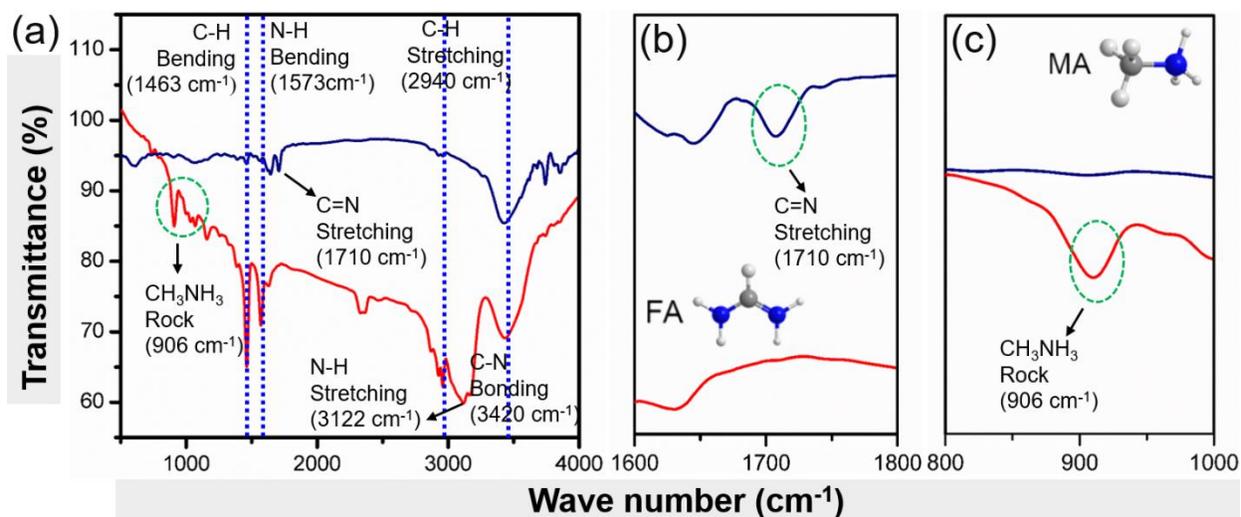


Figure 4. (a) FTIR spectra of fresh $(\text{BA})_2\text{FAPb}_2\text{I}_7$ (blue curve) and $(\text{BA})_2\text{MAPb}_2\text{I}_7$ (red curve) crystals in an open atmosphere. The formation of characteristics $-\text{C}=\text{N}$ stretching band at 1710 cm^{-1} observed for the FA, is absent in MA (b) and the band at 906 cm^{-1} correspondings to CH_3NH_3 rocking is absent for FA (c).

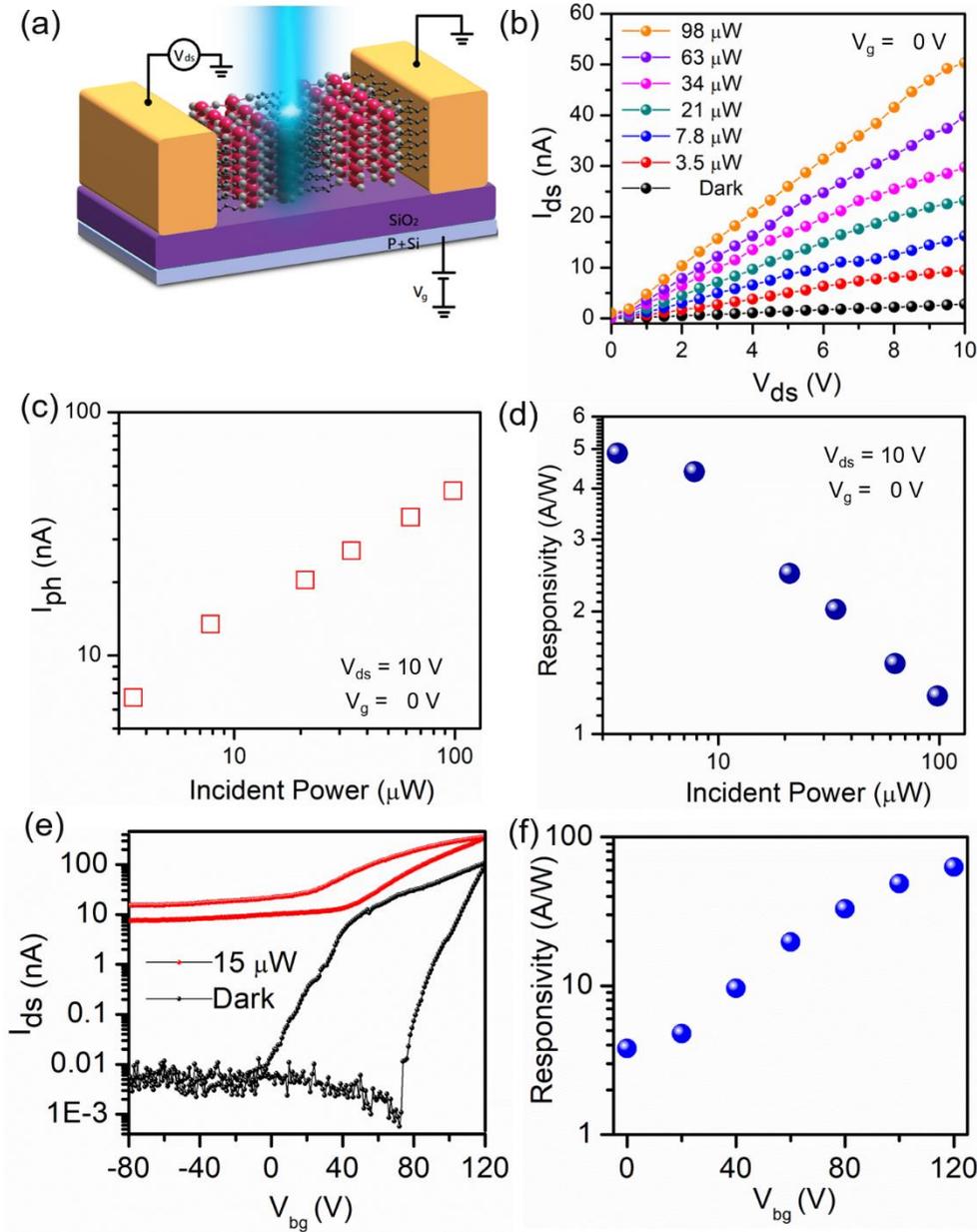


Figure 5. (a) Schematic illustration of the $(\text{BA})_2\text{FAPb}_2\text{I}_7$ -FET device. (b) I_{ds} - V_{ds} curves in the dark and under laser ($\lambda = 488$ nm) irradiation with different intensities. (c) Irradiance dependence of photocurrent of the device under 10 V bias and $V_g = 0$. (d) Responsivity is a function of illumination intensity ranges from 3.5 to 98 μW of 488 nm wavelength. (e) The I_{ds} - V_{bg} curve with V_{bg} scanned from -80 to +120 V at $V_{ds} = 10$ V under light illumination and dark states. (f) R_λ of the $(\text{BA})_2\text{FAPb}_2\text{I}_7$ photodetector as a function of V_{bg} measured at $V_{ds} = 10$ V and $P = 15$ μW .

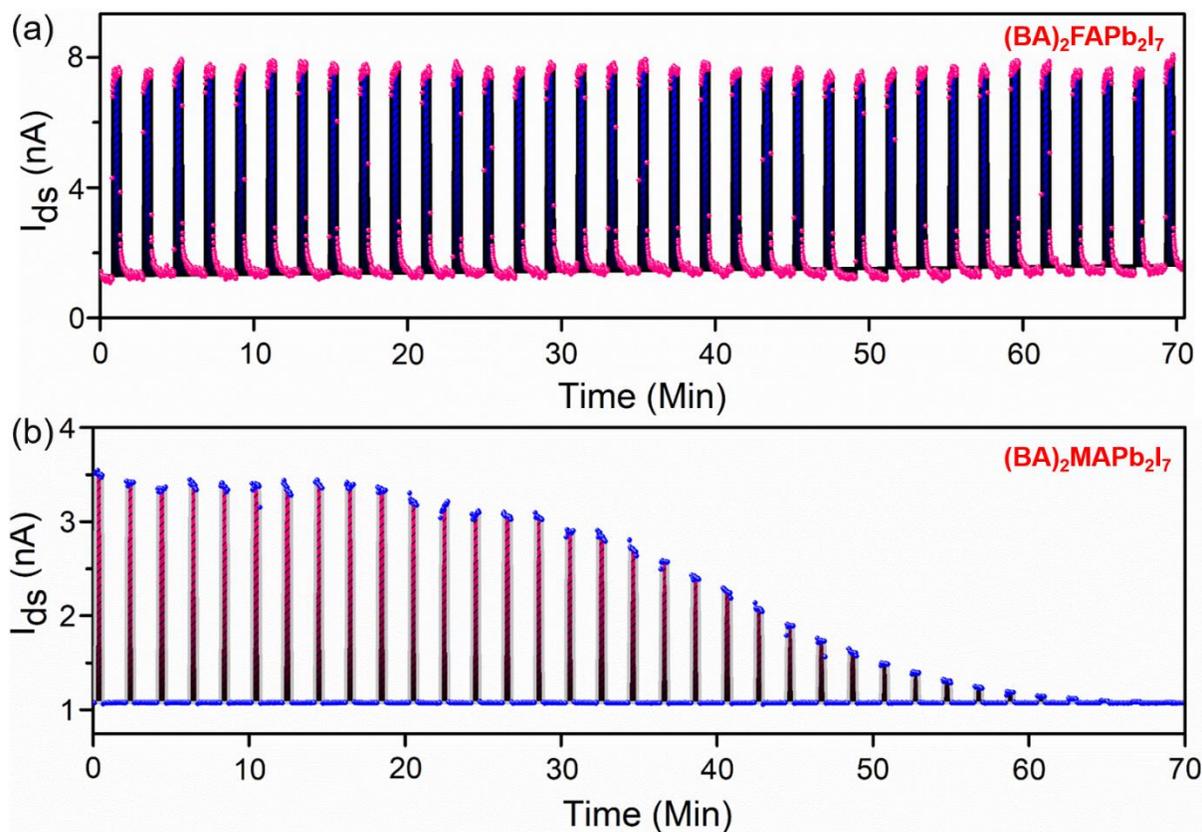


Figure 6. Photoswitching stability study for the 2D $(BA)_2FAPb_2I_7/(BA)_2MAPb_2I_7$ -FET device in response to a train of pulsed illumination at $\lambda = 488$ nm, $P = 3.50 \mu W$, and $V_{ds} = 10$ V at ambient environment. (a) The time-resolved (I_{ds} -t) measurement of $(BA)_2FAPb_2I_7$ shows the strong ambient and laser stability without photocurrent degradation from its initial value at 0 to 60 min long run. (b) The I_{ds} -t curve of $(BA)_2MAPb_2I_7$ starts to degrade after reaching 20 minutes and completely vanishes at 60 minutes.

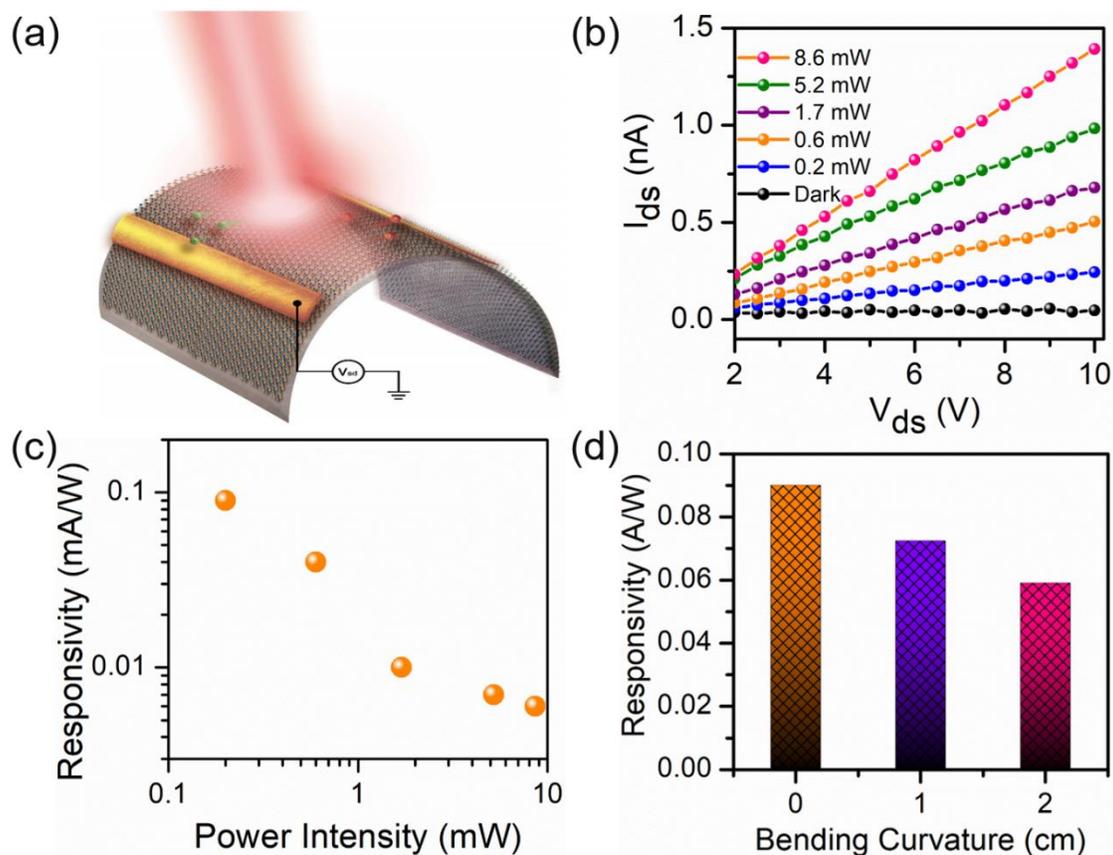
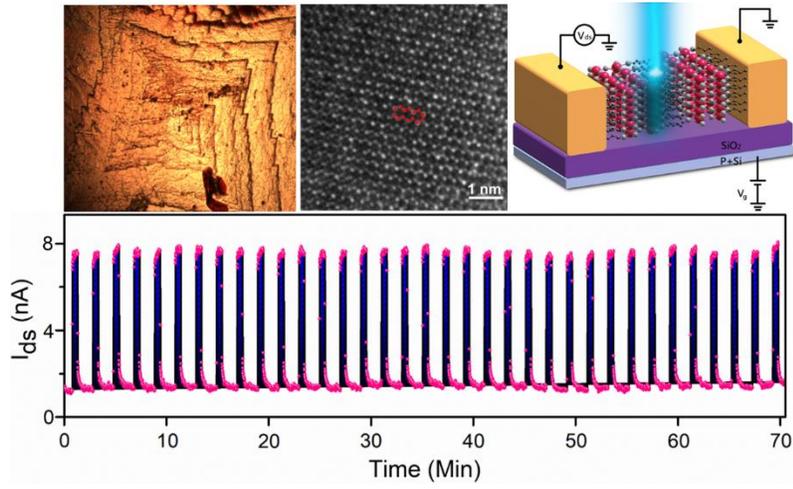


Figure 7. Optoelectronic properties of a flexible photodetector. (a) Schematic illustration of a 2D $(\text{BA})_2\text{FAPb}_2\text{I}_7$ on a PET substrate. (b) The I_{ds} - V_{ds} curve of the few-layered $(\text{BA})_2\text{FAPb}_2\text{I}_7$ photodetector on PET was measured in the dark and light illumination. (c) The calculated responsivity as a function of the illumination intensity acquired in planar geometry at $V_{ds} = 10$ V. (d) Responsivity test of the $(\text{BA})_2\text{FAPb}_2\text{I}_7$ device under bent of different radius (1 and 2 cm) and unbent state.

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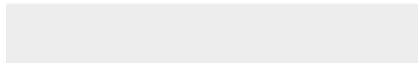
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