

PHYSICAL PRETREATMENT OF BIOGENIC-RICH TROMMEL FINES FOR FAST PYROLYSIS

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ABSTRACT: Energy from Waste (EfW) technologies such as fluidized bed fast pyrolysis, are beneficial for both energy generation and waste management. Such technologies however, face significant challenges due to the heterogeneous nature, particularly the high ash contents, of some municipal solid waste types e.g. trommel fines. A study of the physical/mechanical and thermal characteristics of these complex wastes is important for two main reasons; a) to inform the design and operation of pyrolysis systems to handle the characteristics of such waste; b) to control/modify the characteristics of the waste to fit with existing EfW technologies via appropriate feedstock preparation methods. In this study, the preparation and detailed characterization of a sample of biogenic-rich trommel fines has been carried out with a view to making the feedstock suitable for fast pyrolysis based on an existing fluidized bed reactor. Results indicate that control of feed particle size was very important to prevent problems of dust entrainment in the fluidizing gas as well as to prevent feeder hardware problems caused by large stones and aggregates. After physical separation and size reduction, nearly 70 wt.% of the trommel fines was obtained within the size range suitable for energy recovery using an existing fast pyrolysis system. This pyrolyzable fraction has similar thermochemical properties and could account for about 83% of the energy content of the 'as received' trommel fines sample. Thus indicating that suitably prepared trommel fines samples can be used for energy recovery, with more than 50% reduction in mass and volume of the original waste. Consequently, this can lead to more than 90% reduction in the present costs of disposal of trommel fines in landfills. In addition, the recovered plastics and textile materials could be used as refuse derived fuel.

KEYWORDS: *Trommel fines, biogenic municipal waste, physical pretreatment, fluidized bed reactor*

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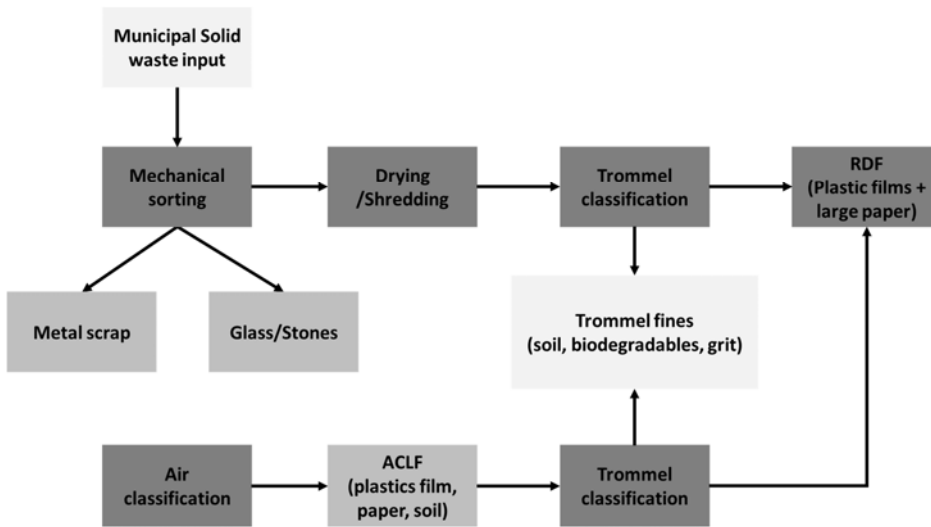
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46 **1. INTRODUCTION**

47 Trommel screens are commonly used in municipal solid waste (MSW) treatment plants. They
48 can be used for both raw MSW and the air-classified light fractions (ACLF) of wastes to separate
49 solid materials into different sizes (Glaub, Jones and Savage, 1982; Kim, Matsuto and Tanaka,
50 2003; Zhang and Banks, 2013). In the case of raw MSW, mechanical sorting and removal of
51 glass, stones and aggregates are first carried out, then the waste is shredded and delivered for
52 size classification on trommel screens. Fine materials, including soil, grit, and much of the
53 biodegradable waste, fall through the screen as "unders", while plastic films and large paper
54 products are retained on the screen as "overs" and may be recycled or marketed as refuse
55 derived fuel (RDF) for energy recovery (Fitzgerald, 2013; Pitchell, 2014). For ACLF, which
56 usually contains plastic film, paper and fine materials such as soil and grit, trommel screens are
57 used to remove the fine fractions of ACLF in order to improve its recyclability or enhance its
58 calorific value for use as RDF (Glaub, Jones and Savage, 1982; Kim, Matsuto and Tanaka, 2003;
59 Zhang and Banks, 2013). Whether applied to raw MSW or ACLF, trommel screens often produce
60 a by-product of mixed compositions, called trommel fines from the mechanical recycling of MSW.
61 A schematic for MSW processing to obtain RDF which generates trommel fines is shown in
62 Figure 1.

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Figure 1. Mechanical recycling of MSW [Fitzgerald, 2013; Pitchell, 2014]

In general, trommel fines are made up of various materials that are less than 70 mm in size and contain both organic and inorganic components (Fitzgerald, 2013; Pitchell, 2014). The compositions of trommel fines thus depend on the initial type and composition of MSW, the severity of mechanical processing of the MSW and the design of the trommel screens. In general, the organic components of trommel fines include fibre, plastics, wood, food waste and textiles, with high calorific values and therefore can be used as an energy resource. The inorganic content of trommel fines includes inert materials such as stones, aggregates, glass

75 and soil.

76 Trommel fines are now included in the UK's Landfill Tax (Qualifying Fines) Order 2015, and
77 this requires landfill operators to conduct loss of ignition (LOI) tests to determine the correct tax
78 liability for trommel fines; £2.65 per tonne dry basis for 'less polluting' waste with <10% LOI, or
79 £86.10 per tonne, (the current higher rate) for waste exceeding 10% LOI. For most operators
80 this means landfill tax for this type of waste will increase. For example, an operator producing
81 40 tonnes per day of trommel fines (dry basis) could see their landfill tax bill increase from about
82 £40,000 to over £1,000,000 per year if it contains more than 10% LOI material (HMRC, 2014;
83 Watts, 2016). Therefore the handling and disposal of trommel fines is now a major problem for
84 the waste management industry, which requires further research. Some proposed methods
85 include further separation and classification of components of trommel fines, for example, to
86 recover smaller fractions of plastics and paper for RDF. This may be an expensive option and
87 the physical separation of inorganics and organics on the basis of size and mass may no longer
88 be feasible at smaller (< 2 mm) scales. Since the the volatile matter content of trommel fines is
89 one of the main concerns for their disposal via landfill (HMRC, 2014; Watts, 2016), technologies
90 are required to make the composition of trommel fines suitable for processing for operators and
91 regulators.

92 Energy from Waste (EfW) technologies may therefore be a viable option for the handling of
93 trommel fines as they can deliver significant benefits in terms of renewable energy recovery and
94 reducing trommel fines to inert materials, which can be landfilled or used in the construction
95 industry. In addition, EfW recovery processes can prevent some of the negative impacts of
96 greenhouse gas emissions and land use issues that are associated with waste landfills. The
97 energy generated from EfW processes can reduce the dependence on fossil fuel for power
98 generation and contribute towards meeting UK renewable energy targets (Hackett et al., 2014).
99 The term 'Energy from Waste' covers a range of processes, both thermochemical and biological,
100 which recover value from waste in the form of energy. Thermochemical processes are usually
101 faster and more robust than biological processes for mixed waste streams. For example,
102 thermochemical processing can handle both biodegradable and non-biodegradable waste,
103 whereas biological processing can only handle readily biodegradable waste (McKendary, 2002).

104 Some thermochemical processes extract the energy directly as heat (incineration) whereas
105 others convert waste into different types of fuel for later use (pyrolysis and gasification). Fast
106 pyrolysis offers an effective and sustainable technology to enable conversion of difficult-to-
107 process solid wastes such as trommel fines due to their heterogenous composition thereby
108 diverting such wastes from landfill. Among the different pyrolysis reactors, fluidized bed reactors
109 can be operated to handle the fast pyrolysis of waste streams with high inorganic contents due
110 to their reliability and ease to operate. Also they are quite simple to scale up from lab to
111 commercial plant scale. However, the characteristics of the inert materials, particularly the
112 particle size, brittleness and hardness, need to be considered in order to minimize the adverse
113 effects on the pyrolysis process and products. Fine dust particles of less than 50 µm may be
114 transported as dust out of the reactor by the fluidizing media and contaminate the pyrolysis
115 products and potentially cause system blockage. In addition, the presence of stones, aggregates
116 and glass can result in significant mechanical hardware problems, such as abrasive wear and
117 tear of reactor interiors as well as blocking of moving parts.

118 During fluidised bed fast pyrolysis, more than 90% of the heat requirement for a feedstock
119 particle is achieved via conduction through contact with the fluidized bed materials, (Bridgwater
120 et. al 1999). Common bed materials include silica sand, which provides a constant temperature
121 distribution within the reactor due to its very efficient heat transferability resulting from its high
122 solid density. A fluidized bed reactor achieves the high heating rates at low residence times for
123 complete thermal degradation by using small feedstock particle sizes (Bilbao et. al., 1994).
124 Feedstocks used in a bubbling fluidised bed (BFB) fast pyrolysis reactor must be prepared to

125 certain specifications. In most pilot and lab scale units, the particle size of feedstock must be
126 between 0.25 - 2.00 mm for effective heat transfer and have been dried to below 10% moisture
127 content (Bridgwater et al., 2000; Chen et. al., 2014). In addition, ash content of feedstock must
128 be minimized as deposition and addition of ash to the bed material will lead to problems in
129 fluidization as well as in heat transfer. However, high ash samples may be used in a fluidized
130 bed reactor with adequate design considerations. The pretreatment of solid waste is also
131 important for reducing formations and emissions of trace levels of toxic organic pollutants, such
132 as dioxins and dioxin-like compounds during pyrolysis process (Guorui et al., 2016; Rong et al.,
133 2016; Yuyang et al., 2017).

134 This study describes a series of procedures for the pretreatment, characterization and
135 preparation of a sample of biogenic-rich trommel fines for fast pyrolysis in a 300 g h⁻¹ BFB
136 reactor. It also investigates the influence of particle size differentiation on sample
137 characterization and thermal degradation properties. Detailed characterization in terms of
138 calorific value and ash contents as well as particle size differentiation of the inert materials are
139 needed to design a suitable fast pyrolysis process for this type of waste with highly variable
140 carbonaceous composition and high contents of ash and inerts.

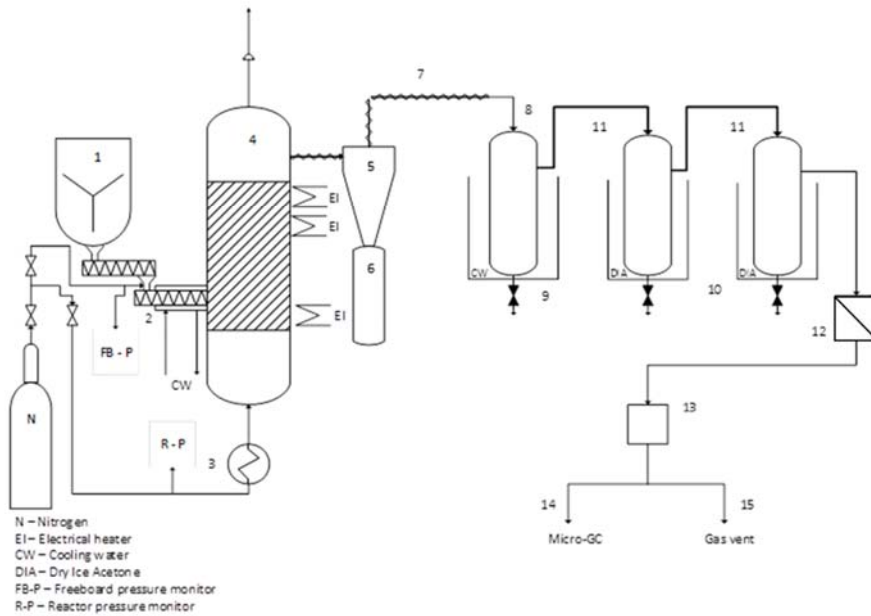
141 **2. MATERIAL AND METHODS**

142 **2.1 Trommel fines feedstock**

143 The sample used in this study was supplied by a UK commercial waste management
144 company – Biffa Ltd from Leicester. The sample supplied was household waste after mechanical
145 removal of the majority of the metals, glass, and plastics material, etc. Then the material was
146 shredded to small particles and processed through a 10 mm rotary drum trommel screen to
147 reduce the inhomogeneity of the sample. On the basis of size range, this feedstock has been
148 classified as trommel fines (Fitzgerald, 2013; Pitchell, 2014).

149 **2.2 Feedstock preparation**

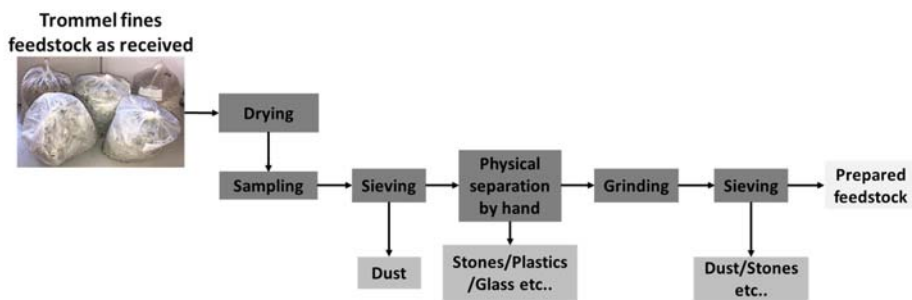
150 The feedstock was prepared based on the feeding requirement of a 300 g h⁻¹ bubbling
151 fluidised bed fast pyrolysis reactor located at European Bioenergy Research Institute (EBRI),
152 Aston University, UK. The reactor consisted of a twin metering screw and a fast screw feeder
153 and usually requires feed particle size range of between 0.25 – 2 mm, with an operating
154 temperature range of 400 – 550°C. The schematic of the BFB reactor is presented in Figure 2,
155 which shows the main process units. To ensure that the trommel fines sample met the particle
156 size requirements of the fluidized bed reactor, the schematic flow diagram in Figure 3 was
157 designed to prepare the sample for fast pyrolysis.
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Figure 2. 300g/h bubbling fluidised bed fast pyrolysis rig set-up.

1 - feed hopper with twin screw feeder, 2 - fast screw, 3 - nitrogen preheater, 4 - fluidised bed reactor, 5 - cyclone, 6 - charpot, 7 - metal transition pipe, 8 - glass transition pipe, 9 - water cooled condenser, 10 - dry ice acetone condenser, 11 - rubber transition pipe, 12 - cotton wool filter, 13 - gas meter, 14 - micro-gc, 15 - gas vent



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Figure 3. Block flow diagram for trommel fines feedstock preparation

175 2.2.1. Sampling

176 The coning and quartering method (Gerlach et al., 2002) was used to obtain a 3 kg
177 representatative batch of the raw trommel fines sample and all the analyses were based on this
178 sample batch. The method, which is used for sampling large quantities of material, consists of
179 pouring the dried trommel fines into a conical heap upon a solid surface and dividing the heap
180 by a cross. The two opposite corners are taken as the sample and the other two set aside.

181 2.2.2. Sieve analysis

182 Prior to sieving, the feedstock was dried in the oven at 60°C for 24 h. The sieve analysis was
183 carried out according to ASTM D 422 standards using a set of sieves of different aperture sizes
184 (> 3.5 mm) and Powermatic Sieve Shaker in order to separate the desired particle size (0.25 –
185 2mm) for pyrolysis experiments (ASTM, 2007°).

186 2.2.3 Grinding

187 The dried feedstock was ground with a Retsch Ltd., Germany, Heavy-Duty Cutting Mill, Type
188 SM2000 with interchangeable screens of varying sizes (1-10 mm). The choice of screen was
189 based on the feeding requirement of the fast pyrolysis reactor which are limited to <2 mm to
190 prevent blockages and feeding problems during fast pyrolysis experiments and to bring the
191 feedstock particles to the optimum pyrolysis temperature and minimize exposure to lower
192 temperatures which favour the formation of char (Bridgwater, 2012).

193 2.2.4 Physical separation

194 Preliminary grinding test of the feedstock proved a challenge as the inorganics, textiles and
195 plastics caused a blockage in the grinding mill (Supplementary Information S11). To aid the
196 grinding, process a physical separation step was added to the feedstock preparation steps, which
197 involved the manual removal of visible glass, stones, metal and plastics from the feedstock
198 fraction with size range > 2 mm (Supplementary Information S12). This step did not remove the
199 entire inorganics and plastics present, but was found to be beneficial for the grinding process.
200 Although this step can be achieved for a laboratory scale process, a more suitable method such
201 as mechanical sorting would be needed for an industrial scale process in order to recover smaller
202 fractions of plastics (< 10mm) for RDF and to reduce the abrasive wear and tear of reactor
203 interiors as well as blocking of moving parts with respect to the twin and fast screw feeder of the
204 fast pyrolysis system. For example, a water floatation process has been trialled, which led to the
205 settling of heavy fractions (stones, sand and aggregates), with the suspension of finer relevant
206 fractions. Details and results of this trial will be published later.

207 2.3 Physico-chemical analysis of trommel fines and fractions

208 The as recieved trommel fines samples and the different size range fractions from the sample
209 preparation procedures were analysed to determine their thermochemical properties. Due to the
210 heterogenous nature of the sample, the coning and quartering method (Gerlach et al., 2002)
211 was used to obtain 100 g batch samples from the feedstock particle size fractions of <0.25 mm,
212 0.25 – 2 mm, 2 – 3.5 mm and 3.5 mm above after grinding and sieving to minimize deviations.
213 Five replicate analyses were also carried out to account for sample varaiation, for which
214 standard deviations were derived.

215 2.3.1. Proximate analysis

216 The proximate analysis of samples involved the determination of the moisture, ash, volatile
217 matter and fixed carbon contents.

218 The moisture contents were determined according to ASTM E1756-01 principles on a dry
219 basis (ASTM, 2007b). The percentage weight loss on a dry weight basis of a pre-weighed
220 sample heated at 105°C to constant weight was recorded. Using the cone and quartering method,
221 five aluminium boats with 4 – 5 g of the trommel fines sample were placed in an oven at 105°C
222 for 6 h. The samples were then cooled in desiccators and re-weighed. The process was repeated
223 hourly for each sample until constant weight was achieved.

224 The ash contents were determined according to the ASTM E1755-01 method (ASTM, 2007c).
225 Five crucibles and lids were put in a Carbolite AAF1100 furnace and heated to 900°C for 3 h;
226 crucibles were then removed from the furnace and cooled in a desiccator. The crucible weights
227 were recorded and then approximately 4 - 5 g of dried feedstock was weighed into each crucible.
228 The crucibles and samples with their lids placed at an angle were placed in a furnace and heated
229 to 250°C at 10°C min⁻¹ and held for 30 min, then increased to 575°C for 5 h. After 5 h, the
230 crucibles were removed and cooled in a desiccator for one hour. Each crucible was weighed to
231 the nearest 0.1 g. Crucibles were replaced in a furnace and heated at 575°C for 1 h periods until
232 the crucible weigh was constant to within 0.3 g. The ash content was then obtained by dividing
233 the weight of the samples by the pre-drying weight and expressed as a percentage of the original
234 weight. The average of five samples was taken to further reduce the deviation.

235 Volatile matter was obtained on a moisture free basis. Volatile matter is the weight loss
236 resulting from heating the sample under controlled conditions. The volatile content of the sample
237 is taken as the weight loss at 950°C (ASTM D3175-89) for 7 min (ASTM, 1997). Five crucible
238 weights were recorded and then approximately 4 to 5 g of dried feedstock was weighed into
239 each crucible. The crucibles and samples with their lids fully sealed were placed in a Carbolite
240 AAF1100 furnace and heated to 950°C and held for 7 min. Upon completion the oven was turned
241 off the crucibles were removed and cooled in a desiccator. Each crucible was weighed, and the
242 average of five samples was taken to further reduce the deviation. The fixed carbon value was
243 obtained by difference.

244 2.3.2. Ultimate analysis

245 A Carlo-Erba 1108 elemental analyser was used to determine the elemental composition of
246 a sample in terms of carbon, hydrogen and nitrogen. Other elements sometimes included in the
247 analysis include, sulphur and oxygen; oxygen is often determined by percentage difference
248 (Aiken, DeCarlo and Jimenez, 2007). Using the cone and quartering method, five different 5 g
249 samples were dried in the oven at 60°C for 24 h prior to analysis and were ground using a coffee
250 grinder. The cone and quartering method was used again to obtain about 1 g from each of the
251 grounded samples for analysis. The results are reported on a dry basis to avoid reporting
252 moisture as additional hydrogen and oxygen (Stahl et al., 2004).

253 2.3.3. Calorimetry

254 The bomb calorimeter experiment is the standard method (ASTM D2015) used to determine
255 the higher heating value for a sample (ASTM, 2000). Results obtained from the bomb calorimeter
256 experiment indicate the samples higher heating value (HHV). Using the cone and quartering
257 method five different 5 g samples were dried in the oven at 60°C for 24 h prior to analysis. The
258 cone and quartering method was used again to obtain approximately 1 g each from the 5
259 different samples, which was burnt completely in an excess oxygen environment in a steel
260 vessel, which is called a bomb using a Parr 6100 calorimeter. The reaction takes place at
261 constant volume.

262 In addition, a mathematical equation (Eq. 1) correlated from the proximate analysis of

263 different biomass from the literature (Parikh, Channiwala and Gosal, 2004) was used to calculate
264 the heating value on a dry basis. The calculated and experimental results would be displayed
265 for comparisons;

$$266 \quad \text{HHV}_{(\text{dry})} = 0.3536 \text{ FC} + 0.1559 \text{ VM} - 0.0078 \text{ A} \quad (\text{Eq. 1})$$

268
269 Where FC is fixed carbon, VM is volatile matter, and A is ash content.

270 2.3.4. Thermogravimetric analysis (TGA)

271 To study pyrolysis under dynamic heating for the trommel fines feedstock, a PerkinElmer
272 Pyris 1 thermogravimetric analyser was used. A pyrolysis heating rate of $10^{\circ}\text{C min}^{-1}$ was used
273 and heating from ambient temperature to 550°C in nitrogen flow of 30 ml min^{-1} . Using the cone
274 and quartering method, five different 5 g samples were dried in the oven at 60°C for 24 h prior
275 to analysis and were ground using a coffee grinder. The cone and quartering method was used
276 again to obtain about 2 -3 mg from each of the five different grounded samples which was
277 placed on a ceramic crucible on the analyser tray. The sample crucible was placed in a sensitive
278 thermo-balance. The sample was subjected to heat from an external furnace to pre-set
279 temperatures and heating rate. The weight loss as a result of thermal degradation was measured
280 and recorded on the program software. All pyrolysis TGA experiments were conducted in an
281 inert atmosphere of nitrogen. Each analysis was repeated 5 times.

282 2.4. Cold feeding trial for fast pyrolysis

283 After the sample preparation steps described above, the size fraction for fast pyrolysis was
284 cold-fed into an existing bubbling fluidized bed reactor to monitor particle behaviour prior to
285 actual pyrolysis tests. The feeders attached to the fluidised bed reactor consist of an air tight
286 hopper with nitrogen purge with a Ktron KT-20 gravimetric speed regulated twin metering screws
287 attached to a high speed feed screw, which is water cooled at the feed point to minimise pre-
288 pyrolysis (see Figure 2). The feedstock feed rate can be adjusted using the computing system
289 on the Ktron KT-20 gravimetric feeding system to adjust the speed of the feeding twin screw in
290 to the fast screw; the set feed rate is then displayed on the system's LED screen.

291 3. RESULTS AND DISCUSSION

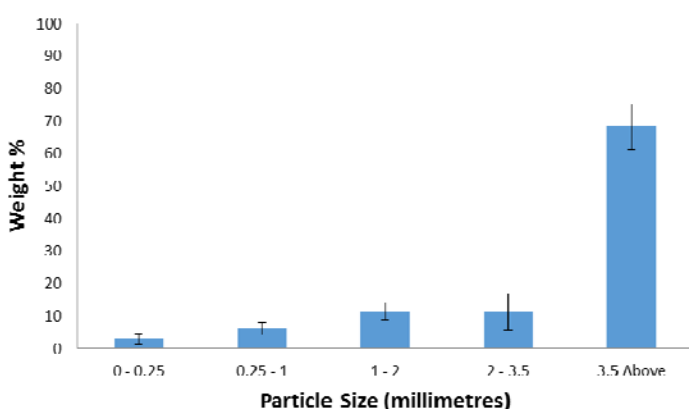
292 3.1 Particle size distribution in the trommel fines sample

293 Figure 4 shows the average particle size distribution by weight percent for five replicate 0.5
294 kg samples of the dried trommel fines feedstock before any sample preparation (grinding and
295 separation). The results shows that only about 17.6 wt. % of the feedstock was initially within the
296 size range (0.25 – 2 mm) that could be used for fast pyrolysis with an existing BFB. The 17.6
297 wt% comprised of fractions with particle sizes of 1 - 2 mm range accounting for an average of
298 11.5 wt% and 0.25 - 1 mm size range, which accounted for 6.13 wt% of the total weight of the
299 trommel fines sample, respectively. Visual observation (Supplementary Information SI3)
300 showed that these two size ranges were similarly made up of few tiny pieces of plastics and
301 glasses, with the bulk of the fraction consisting of wood, paper/cardboard, textile material and
302 sand; therefore could be suitable for energy recovery by fast pyrolysis.

303 A small fraction, accounting for 2.8 wt% was obtained with $<0.25 \text{ mm}$ particle size range and
304 classified as dust, while 79.5 wt. % needed to undergo further processing. On average, the
305 fraction with particle sizes $>3.5 \text{ mm}$ accounted for 68.2 wt% of the trommel fines. This size range
306 comprised of mostly paper pellets, plastics, glass, stones, bones, miscellaneous wood, textile

307 and metals. In addition, the fractions with particle sizes >2mm would require size reduction to
308 enable feeding into the pyrolysis equipment. However, the presence of inert materials such as
309 stones and glass would cause mechanical problems for the pyrolysis system.

310 Taken together, the two fractions with particle sizes >2 mm were combined and processed
311 as follows; after physically removing the visible stones, glasses and plastics by hand, the
312 remainder was ground and sieved to achieve a fraction with particle size range suitable for the
313 fast pyrolysis equipment. For a large batch of waste the size range classification can be achieved
314 using an industrial sieve but further research is required for appropriate method to remove the
315 inorganics in a large batch of waste.
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319 Figure 4. Average particle size distribution of dried trommel fines sample before preparation (5 replicates)
320

321 3.2 Characterization of trommel fines in relation to particle size distribution

322 Table 1 shows the results of the proximate analysis of the trommel fines feedstock in different
323 size ranges on a dry basis. The data in the table also reports the calculated HHV as well as the
324 average of those obtained using a bomb calorimeter.

325 The proximate analysis data are relevant in determining the suitable quantity and thus the
326 feeding rate for fast pyrolysis processing, which depends on the proportion of volatile matter in
327 the feedstock and the rate of its thermal degradation. Also, the analysis offers a preview on the
328 mass balance of the system.

329 The moisture content of the feedstock prior to processing was found to be 46.0 ± 3.23 wt.%
330 on a dry basis, therefore requiring drying before processing to aid the grinding and sieving
331 process. After oven drying at 60 °C the moisture content decreased to less than 13.5 wt.% for
332 all fractions. However, the 2 – 3.5 mm fraction was found to retain the most moisture, possibly
333 due to its high organic content, which enhanced moisture retention. The moisture in the trommel
334 fines feedstock will eventually end up in the pyrolysis products, mainly in the liquid product. This
335 is because the moisture in the feed must be evaporated before the thermal degradation of the
336 particle will begin. The presence of water has also been shown to cause secondary reactions in
337 some cases (He et al., 2009; Westerhof et al., 2007; Maniatis 1988; Gray et al., 1985). As water
338 is an unwanted compound in most bio-oil, due to its lowering effect on the heating values, a
339 moderate amount of moisture is known to impact positively on the viscosity of the oil product.
340 Bridgwater et al. (2000) recommends that the moisture content for biomass for fast pyrolysis

341 processes be around 10 % and this would be applicable to a biogenic-rich waste feedstock.
 342 Therefore, further drying may be required before the prepared feedstock could be used;
 343 however, this can be achieved via combustion of dusty fraction and char in the solid residues.

344 There was a clear indication of variation of ash content between the size fractions as seen in
 345 Table 1. The ash content decreased with increasing size fraction. The ash content of all the
 346 samples were between 31 – 54 wt.% on a dry ash free basis and shows significant difference
 347 between the size ranges, in particular, the particle size range of (0.25 – 2 mm) deemed for fast
 348 pyrolysis has an ash content of 42.1 ± 3.41 wt.% which is slightly similar to the as received
 349 sample. The inverse pattern was observed with the volatile content and the volatile content
 350 increased with particle size. There was no apparent trend with the fixed carbon with respect to
 351 particle size. From these results there was an indication that different fractions of the trommel
 352 fines can have significant differences in chemical properties. The high ash content in the feed,
 353 especially in the size fraction of 0.25 – 2 mm for the fast pyrolysis process was obviously due to
 354 the presence of high amounts of fine inorganics as the ash content is known to be dependent
 355 on the inorganic components of the feedstock. The high inorganics in the feed could arise from
 356 a number of reasons, such as the source of the feedstock and the technique used to separate
 357 the waste streams. The high ash content will lead to an increase in volume and weight of bed
 358 material in a fluidized bed pyrolysis process. Research has shown that inorganic compounds
 359 present in a feedstock promote the formation of char and gas at the expense of pyrolysis liquid
 360 yield. An increase in char and gas yield at the expense of bio-oil due to the presence of ash
 361 during pyrolysis was observed in a number of studies (Hodgson et al., 2010; Teng et al., 1998;
 362 Varhegyi et al., 1989; Sekiguchi et al., 1984). The reduced volatiles may be due to the size
 363 reduction process; for instance, the size reduction process (Figure 6) might have aided the
 364 degradation of the feedstock as well as eliminating certain materials (rubber, textile, and plastics)
 365 that would have improved the volatile content of the feed. The reduced volatiles in this feedstock
 366 were an early indicator of low liquid and gas yields from fast pyrolysis.

367 The experimental heating values for all the samples were between $7.8 - 13.2$ MJ kg⁻¹, on a
 368 dry basis with the heating value increasing with increasing size fraction. This was an indication
 369 that pyrolysis liquids and bio-fuels with moderately high-energy content may be obtained from
 370 this feedstock. The experimental results compared well with the theoretical heating values (Table
 371 1) as they were observed to be similar and increasing with size fraction.

373 Table 1. Results of average proximate analyses and heating values of trommel fines in relation to size
 374 ranges. (5 replicates)

ANALYSIS	UNITS	As received	< 0.25 (mm) ^a	0.25 – 2 (mm) ^{a, b}	2 – 3.5 (mm) ^a	3.5 Above (mm) ^a
Ash content ^a	wt.%	43.3 ± 4.81	53.8 ± 6.20	42.1 ± 3.41	34.9 ± 3.56	31.5 ± 2.08
Volatile Matter ^a	wt.%	46.6 ± 3.53	40.0 ± 3.03	49.7 ± 6.50	56.7 ± 3.28	60.1 ± 4.65
Fixed Carbon ^c	wt.%	11.1	6.20	8.15	8.39	8.38
Moisture content	wt.%	46.0 ± 3.23	9.56 ± 1.01	12.5 ± 3.04	13.2 ± 0.75	12.5 ± 0.92
Bomb Calorimeter ^a	MJ kg ⁻¹	11.6 ± 2.59	7.78 ± 0.87	11.8 ± 0.27	13.2 ± 0.59	12.5 ± 0.41
Calculated ^a	MJ kg ⁻¹	10.7	8.01	10.3	11.5	12.1

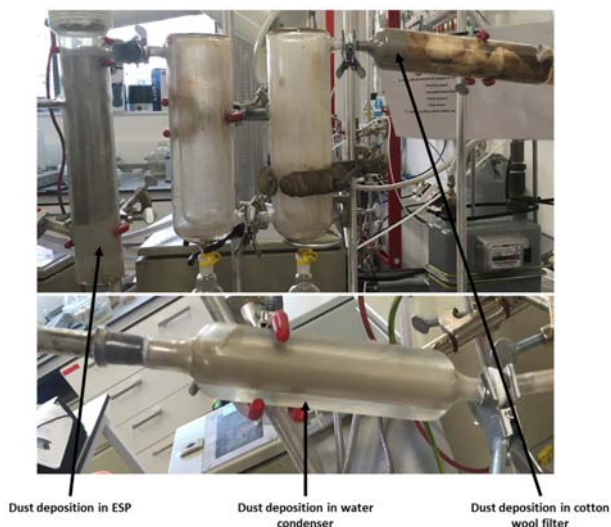
^a dry basis; ^b prepared size fraction for fast pyrolysis experiments, ^c calculated by difference

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377 **3.3. Physical preparation of trommel fines for fast pyrolysis**

378 Due to the limitations of the feeding system (allows only particle sizes < 2 mm) of an existing
379 fast pyrolysis reactor, it was decided to test-feed the reactor with the combined fractions with
380 particle size ranges covering between 0.25 – 2 mm. This initial tests revealed serious problems
381 of dust entrainment and deposition throughout the downstream parts of the pyrolysis system
382 (condensers and filters), as shown in Figure 5. Apparently, the amount of dust (< 0.5 mm)
383 produced during the feeding increased due to the vibration of the fast screw rotating at a pre-
384 set rate of 100 rpm. This dust was easily transported through the rig via the nitrogen gas
385 connected to the feeding system and the fluidizing nitrogen in the reactor. This could easily pose
386 significant problems during the operation of the pyrolysis equipment. Firstly, the dust could settle
387 along narrower pipes and cause blockage which would in turn lead to build up of pressure in the
388 rig. Such pressure build-up could lead to explosion and loss of containment. Secondly the dust,
389 which still contains combustible organic matter may be susceptible to dust explosion with serious
390 consequences. Thirdly, the dust would contaminate the condensable pyrolysis products
391 downstream, cause blockage to the fast screw when feeding and affect the safe operation of
392 the rig. The solution to this challenge was to reduce the content of the fine dust particles by
393 physical separation via sieving. Hence, when the feed particle size was adjusted to 0.5 – 2 mm,
394 the dust deposition and transportation through the rig stopped. Although, this would decrease
395 the fraction obtained for energy recovery via fast pyrolysis, the safety of operation was deemed
396 of much higher importance than a few losses.

397 Hence, this particle size range was considered for the fast pyrolysis of the trommel fines,
398 following the sample preparation protocol depicted in Figure 6. As shown in Figure 6, a 3 kg
399 batch of trommel fines was used directly for the sample preparation involving physical separation
400 by hand, grinding and sieving. Table 2 shows the results of the preparation protocol. After feed
401 preparation, 69.4 wt. % of the 3 kg trommel fines batch was obtained with a suitable particle size
402 range for the fast pyrolysis process (0.5 – 2mm).
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405 Figure 5. Accumulation of dust in downstream parts of a fast pyrolysis rig during feeding trial

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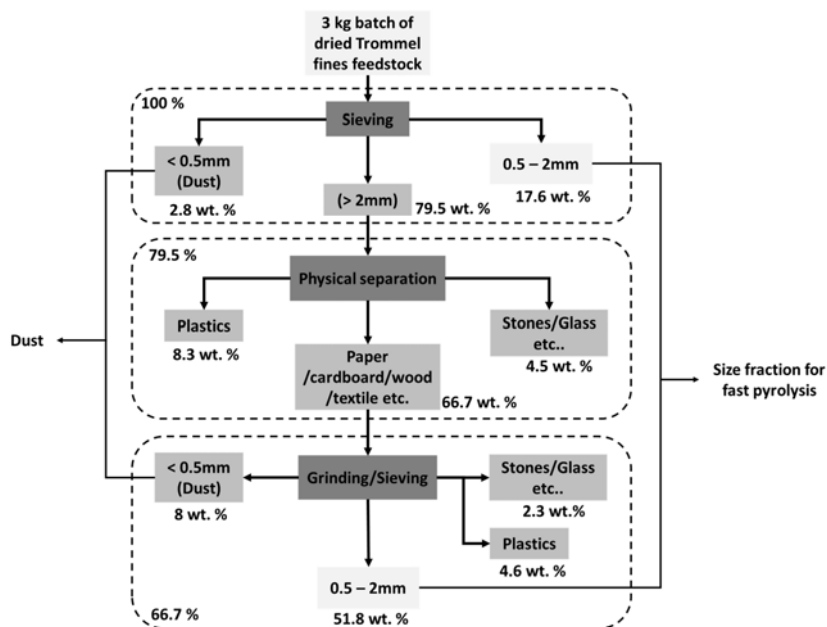


Figure 6. Detailed scheme for trommel fines feedstock preparation

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Table 2. Updated summary of trommel fines main components from feedstock preparation

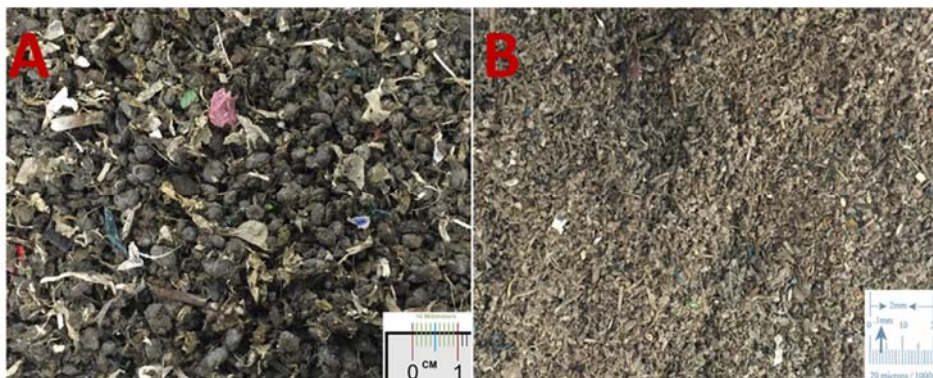
Fractions	Weight %
< 0.5 mm	10.8
0.5 - 2mm*	69.4
Stones/Glass etc.	6.8
Plastics	12.9
TOTAL	99.9

* Size fraction for fast pyrolysis

410

411 An enlarged photographs of the 'as-recieved' sample and the prepared sample fraction
 412 deemed suitable for fast pyrolysis is shown in Figure 7. The calorific value (HHV) of the fraction
 413 for pyrolysis was determined to be $13.8 \pm 0.32 \text{ MJ kg}^{-1}$. This value is slightly higher to the value
 414 for the 0.25 – 2 mm fraction shown on Table 2, indicating that the removed dusty fraction
 415 composed of mostly inert materials. In total, the 0.5 – 2 mm fraction represented 82.6 % of the
 416 energy content of the original batch of trommel fines. Overall, some 10.8 wt. % (<0.5 mm)
 417 comprised of mainly dust (ash), which was eliminated. This fraction could be burnt in an

418 industrial process to provide heat for the pyrolysis process as it had an appreciable heat content
 419 (7.8 MJ kg^{-1}). The physical separation by hand yielded 12.9 wt. % and 6.8 wt.% of plastics and
 420 inorganics (stones, bone, etc.) respectively. On an industrial scale, the plastics and textiles could
 421 be recycled for RDF and while the inorganics (glass and stones) can be used in construction.
 422



423
 424
 425 Figure 7: A. Trommel fines as received (<10 mm size range); B. Prepared trommel fines fraction (0.5 –
 426 2 mm size range) for fast pyrolysis
 427

428 The ultimate analysis, on dry, ash-free basis, of the prepared trommel fines feedstock (size
 429 range 0.5 - 2 mm in Table 3) shows elemental compositions similar to literature data for a refused
 430 derived fuel from municipal solid waste (MSW) as reported by other authors (Efika, Wu and
 431 Williams, 2012; Blanco et al., 2012; Materazzi et al., 2015). The prepared trommel fines
 432 feedstock has a lower carbon content of 34 wt. % when compared to literature data range of 43
 433 - 62.1 wt. % for RDF. The oxygen contents of the feedstock also differ with this feedstock having
 434 lower oxygen content of 13.7 wt.% when compared to literature, which stands at between 26.52
 435 and 37.9 wt.%. These differences could be attributed to the source and composition of the waste.
 436 One clear observation was that the feedstock has a significantly high nitrogen content of 4.79
 437 wt. % when compared with other MSW literature data of 0.1 - 1.82 wt. % (Efika, Wu and Williams,
 438 2012; Blanco et al., 2012; Materazzi et al., 2015). The nitrogen contents also serve as an
 439 indication of the possibility of NO_x compounds forming during the oxidative thermochemical
 440 processing of the feedstock (Diebold, 2002). This is undesirable in terms of environmental
 441 considerations. The feedstock showed sulphur contents like those reported for MSW in literature
 442 (Materazzi et al., 2015).
 443
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 445

Table 3: Ultimate analysis of trommel fines 0.5 - 2 mm size range (5 replicates)

Elements	*Wt.%
Nitrogen	2.75 ± 3.16
Carbon	33.71 ± 6.93
Hydrogen	4.62 ± 0.92
Sulphur	0.26 ± 0.20
Oxygen	17.06 ± 8.21

*remainder was classified as ash

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448 Thermal degradation rate of solid fuels is very important in the design of a fluidized pyrolysis
449 system. Fundamental to the degradation rate are the rates of bond breaking, formation and
450 devolatilisation of small stable molecules. The TGA and DTG curves of the prepared trommel
451 fines feedstock are shown in Figure 8. Three major weight loss steps are evident from the DTG
452 curve. The initial weight loss step, which accounted for the removal of moisture from the
453 feedstock, occurred between 40 and 106°C. The second weight loss step happened between
454 110 and 390 °C. The second step was the major pyrolysis process and the inflection point of this
455 step was at 350 °C and that showed the temperature at which weight loss was at its maximum.
456 This loss can be attributed to the degradation of components such as hemicellulose, cellulose
457 and lignin of the sample. The third weight loss step occurs between 400 and 523°C. The small
458 weight loss with a maximum degradation temperature of about 500°C may be attributed to the
459 small amount of fine plastic particles remaining in the sample. The total weight loss as can be
460 seen on the TGA curve of the trommel fines feedstock stands at about 54% between 40 °C and
461 550°C, which is similar to the upper limit sum of VM and moisture contents of the sample (53.7
462 wt%).

463 Clearly, Figure 8 confirms the high ash content of the prepared trommel fines sample but it
464 also shows that the sample contains a significant proportion of volatile mater from which energy
465 can be obtained via fast pyrolysis. Recovery of this energy will be important to meet landfill
466 disposal requirements in terms of the loss on ignition (LOI) limits and for sustainable waste
467 management. The results of the ash content analysis of the feedstock, suggest that about 50%
468 of the feedstock can be used for energy recovery and diverted from landfill. In addition, the
469 seemingly inert ash product may be used in construction. This will reduce the amount of landfill
470 tax an operator that produces 40 tonnes a day would pay from over £1,000,000 to less than
471 £60,000. This estimate takes into consideration the reduction in mass and volume of waste due
472 to energy recovery as well as the reduced tariff applicable to the landfilling of ash-rich, low LOI-
473 bearing solid residues.

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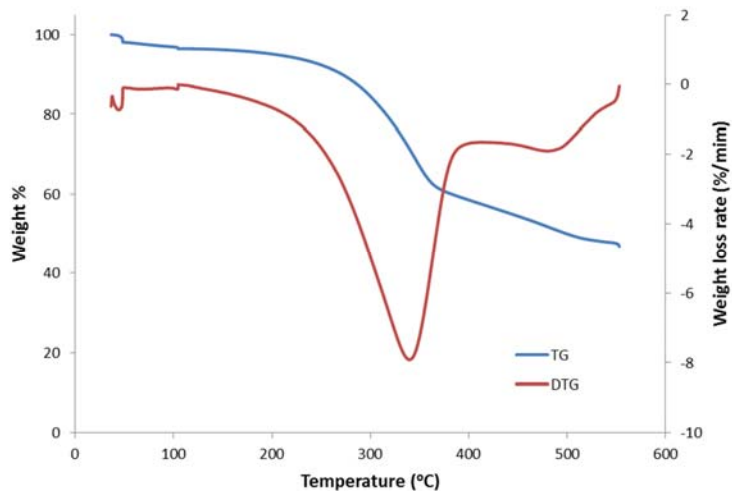


Figure 8. TG and DTG curves of trommel fines

495 **4. CONCLUSIONS**

496 A sample of trommel fines obtained from a UK Waste Mangement company showed a wide
497 range of particle sizes and contained mixtures of different materials from plastics, paper and
498 cardboard to stones and bricks. The increasingly tight regulations for disposal of trommel fines
499 in landfills, especially in terms of LOI limits, have necessitated the need to investigate alternative
500 processes for its management. Thermochemical process via fast pyrolysis looks like a plausible
501 solution, however the physical characteristics of trommel fines would need to be adjusted to meet
502 the requirements of fast pyrolysis. Initial results of proximate analyses showed that different size
503 fractions of trommel fines have differences in properties. Ash, volatile content and heating values
504 varied in relation to the range of particle sizes. However, the removal of glass, metals and inert
505 materials such as stones etc., which do not contribute to the energy content of the waste, is
506 highly recommended to reduce the volume of waste and minimize reactor damage which can be
507 done via physical separation.

508 Although, manual separation appears plausible for a lab scale process, further research is
509 required for inorganic removal pretreatment method appropriate for a large batch of waste. This
510 is a topic for future work. In addition, dust formation is a potential hazard during the feeding for
511 fast pyrolysis of trommel fines, but this can be minimized by using the appropriate particle size
512 range after size reduction via screening, grinding and sieving. This preliminary work suggests
513 that appropriate feedstock preparation is needed to make thermal recovery of energy possible
514 from heterogeneous and complex waste materials such as trommel fines.

515 Energy recovery from trommel fines has a potential to reduce the cost of handling and
516 disposal this heterogenous waste. Fast pyrolysis tests using the fluidized bed reactor will be
517 carried out of the 0.5 – 2 mm fraction in future to investigate its suitability to handle this type of
518 sample. In addition, the possibility of designing a reactor suitable for the pyrolysis of trommel
519 fines would be investigated.

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523 **REFERENCES**

524 ASTM, 2007a. ASTM D422 Standard Test Method for Particle-Size Analysis of Soils. American
525 Society for Testing and Materials International, United States

526
527 ASTM, 2007b. ASTM E1756-01 Standard Test Method for Determination of Total Solids in
528 Biomass. American Society for Testing and Materials International, United States

529
530 ASTM, 2007c. ASTM E1755-01 Standard Test Method for Ash in Biomass. American Society
531 for Testing and Materials International, United States

532
533 ASTM, 2000. ASTM D2015 Standard Test Method for Gross Calorific Value of Coal and Coke
534 by the Adiabatic Bomb Calorimeter. American Society for Testing and Materials International,
535 United States

536
537 ASTM, 1997. ASTM D3175-89a Standard Test Method for Volatile Matter in the Analysis Sample

538 of Coke and Coal. American Society for Testing and Materials International, United States
539
540 Aiken, A.C., DeCarlo, P.F., Jimenez, J.L. 2007. Elemental Analysis of Organic Species with
541 Electron Ionization High-Resolution Mass Spectrometry. *Anal Chem.* 79 (21), 8350 – 8358.
542
543 Bilbao, R., A. Millera, and M. Murillo, B., *Heat transfer and weight loss in the thermal*
544 *decomposition of large wood particles*, in *Advances in Thermochemical Biomass Conversion*,
545 A. Bridgwater, V., Editor. 1994, Blackie Academic & Professional: Glasgow. p. 833-845.
546
547 Blanco, P.H., Wu, C., Onwudili, J.A., Williams, P.T., 2012. Characterization of tar from the
548 pyrolysis/gasification of refuse derived fuel: Influence of process parameters and catalysis.
549 *Energy & Fuels* 26, 2107-2115.
550
551 Bridgwater, A.V., *Review of fast pyrolysis of biomass and product upgrading*. Biomass and
552 Bioenergy, 2012. **38**(0): p. 68-94.
553
554 Bridgwater, A.V. and G.V.C. Peacocke, *Fast pyrolysis processes for biomass*. Renewable and
555 Sustainable Energy Reviews, 2000. **4**(1): p. 1-73.
556
557 Bridgwater, A.V., D. Meier, and D. Radlein, *An overview of fast pyrolysis of biomass*. Organic
558 Geochemistry, 1999. **30**(12): p. 1479-1493.
559
560 Chen D., Yin, L., Wang, H., and He, P., Pyrolysis technologies for municipal solid waste: A
561 review. *Waste Management*, 2014, 34(12),p. 2466-2486
562
563 Diebold, J.P., 2002. A review of the Chemical and Physical Stability of Fast Pyrolysis Bio-Oils.
564 In (ed.) Bridgwater, A.V. *Fast Pyrolysis of Biomass: A Handbook*, 243-292. CPL Scientific
565 Publishing: Newbury.
566
567 Efika, C.E., Wu, C., Williams, P.T., 2012. Syngas production from pyrolysis-catalytic steam
568 reforming of waste biomass in a continuous screw kiln reactor. *J. Anal Appl Pyrol.* 95, 87-94.
569
570 Fitzgerald, G.C., 2013. Pre-processing and treatment of municipal solid waste (MSW) prior to
571 incineration. In. Klinghoffer, N.B. and Castaldi, M.J. (eds). *Waste to Energy Conversion*
572 *Technology*, 55-71. Woodhead Publishing Limited, Cambridge
573
574 Gerlach, R.W., Dobb, D.E., Raab, G.A., Nocerino, J.M., 2002. Gy sampling theory in
575 environmental studies. 1. Assessing soil splitting protocols. *J. Chemometrics*, 16, 321-328
576
577 Glaub, J.C., Jones, D.B., Savage, G.M., 1982. The design and use of trommel screens for
578 processing municipal solid wastes. In (ed) Grillo, L.M., *Proceeding sof the National Waste*
579 *Processing Conference*. Am. Soc Mechanical Engineers May 2-5, New York. Available at:
580 [//www.seas.columbia.edu/earth/wtert/sofos/nawtec/1982-National-Waste-Processing-](http://www.seas.columbia.edu/earth/wtert/sofos/nawtec/1982-National-Waste-Processing-Conference/1982-National-Waste-Processing-Conference-41.pdf)
581 [Conference/1982-National-Waste-Processing-Conference-41.pdf](http://www.seas.columbia.edu/earth/wtert/sofos/nawtec/1982-National-Waste-Processing-Conference/1982-National-Waste-Processing-Conference-41.pdf). Accessed 26 Sep.2016
582
583 Gray, M.R., W.H. Corcoran, and G.R. Gavalas, *Pyrolysis of a wood-derived material. Effects of*
584 *moisture and ash content*. Industrial & Engineering Chemistry Process Design and
585 Development, 1985. **24**(3): p. 646 651.
586
587 Guorui Liu, Lili Yang, Jiayu Zhan, Minghui Zheng, Li Li, Rong Jin, Yuyang Zhao, Mei Wang.

588 Concentrations and patterns of polychlorinated biphenyls at different process stages of cement
589 kilns co-processing waste incinerator fly ash. *Waste Management* 2016, 58: 280-286;
590

591 Hackett C., Durbin, T.D., Welch, W., Pence, J., Williams, R.B., Jenkins, B.R., Salour, D., Aldas,
592 R., 2004. Evaluation of Conversion Technology Processes and Products. A draft report for the
593 California Integrated Waste Management Board, California, US. Available at:
594 [http://energy.ucdavis.edu/files/05-06-2013-2004-evaluation-of-conversion-technology-](http://energy.ucdavis.edu/files/05-06-2013-2004-evaluation-of-conversion-technology-processes-products.pdf)
595 [processes-products.pdf](http://energy.ucdavis.edu/files/05-06-2013-2004-evaluation-of-conversion-technology-processes-products.pdf). [Accessed 28 September, 2016]
596

597 He, R., et al., *Influence of pyrolysis condition on switchgrass bio-oil yield and physicochemical*
598 *properties*. *Bioresource Technology*, 2009. **100**(21): p. 5305-5311.
599

600 HMRC, 2014. Landfill Tax – Liability of waste ‘fines’. Her Majesty Revenue & Customs.
601 [https://www.gov.uk/government/uploads/system/uploads/attachment_data/file/325100/Landfill_](https://www.gov.uk/government/uploads/system/uploads/attachment_data/file/325100/Landfill_tax_condoc.pdf)
602 [tax_condoc.pdf](https://www.gov.uk/government/uploads/system/uploads/attachment_data/file/325100/Landfill_tax_condoc.pdf). [Accessed 13 Aug. 2016].
603

604 Hodgson, E.M., et al., *Miscanthus as a feedstock for fast-pyrolysis: Does agronomic treatment*
605 *affect quality?* *Bioresource Technology*, 2010. **101**(15): p. 6185-6191.
606

607 Kim, S., Matsuto, T. and Tanaka, N., 2003. Evaluation of pre-treatment methods for landfill
608 disposal of residues from municipal solid waste incineration. *Waste Management & Research*,
609 **21**(5), 416-423.
610

611 Maniatis, K.a.B., A., *Fast Pyrolysis of Biomass*, in *Research in Thermochemical Biomass*
612 *Conversion*, A.V.a.K. Bridgwater, J.L, Editor. 1988, Elsevier Science: New York.
613

614 Materazzi, M., Letteri, P., Taylor, R., Chapman, C., 2005. Performance analysis of RDF
615 gasification in a two stage fluidized bed-plasma process. *Waste management*, *In Press*.
616 doi.org/10.1016/j.wasman.2015.06.016
617

618 McKendry P., 2002. Energy production from biomass (Part 2): conversion technologies. *Biores.*
619 *Technol.* **8**, 37-54
620

621 Parikh, J., Channiwala, S., and Gosal, G., 2005. A correlation of calculating HHV from proximate
622 analysis of solid fuels. *Fuel* **84**, 487-494
623

624 Pitchell, J., 2014. Municipal solid waste processing: material recovery facilities. *Waste*
625 *Mnagement Practices: Municipal, Hazardous and Industrial*. 2nd Edition, 165-196. CRC Press,
626 London,
627

628 Rong Jin, Jiayu Zhan, Guorui Liu, Yuyang Zhao, Minghui Zheng. Variations and factors that
629 influence the formation of polychlorinated naphthalenes in cement kilns co-processing solid
630 waste. *Journal of Hazardous Materials*, 2016, 315: 117-125
631

632 Sekiguchi, Y. and F. Shafizadeh, *The effect of inorganic additives on the formation, composition,*
633 *and combustion of cellulosic char*. *Journal of Applied Polymer Science*, 1984. **29**(4): p. 1267
634 1286.
635

636 Stahl, R., Henrich, E., Gehrman, H.J., Vodegel, S. and Koch, M., 2004. Definition of a Standard
637 Biomass. *RENEW—Renewable Fuels for Advanced Power Trains*, 1 – 14.

638
639 Teng, H. and Y.-C. Wei, *Thermogravimetric Studies on the Kinetics of Rice Hull Pyrolysis and*
640 *the Influence of Water Treatment*. Industrial & Engineering Chemistry Research, 1998. **37**(10):
641 p. 3806-3811.
642
643 Varhegyi, G., et al., *Thermogravimetric-mass spectrometric characterization of the thermal*
644 *decomposition of sunflower stem*. Energy & Fuels, 1989. **3**(6): p. 755-760.
645
646 Watts, J., 2016. Trommel Fines: The Journey so far. Proceedings of CIWM North West Centre
647 Open Meeting, 4th March, 2016. Liverpool, UK
648
649 Westerhof, R.J.M., et al., *Controlling the Water Content of Biomass Fast Pyrolysis Oil*. Industrial
650 & Engineering Chemistry Research, 2007. **46**(26): p. 9238-9247.
651
652 Yuyang Zhao, Jiayu Zhan, Guorui Liu, Zhiyuan Ren, Minghui Zheng, Rong Jin, Lili Yang, Mei
653 Wang, Xiaoxu Jiang, Xian Zhang. Field study and theoretical evidence for the profiles and
654 underlying mechanisms of PCDD/F formation in cement kilns co-incinerating municipal solid
655 waste and sewage sludge. *Waste Management* 2017, 61: 337-344;
656
657 Zhang, Y. and Banks, C., 2013. Impact of different particle size distributions on anaerobic
658 digestion of the organic fraction of municipal solid waste. *Waste Management*, 33(2), 297-307.